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A Formal Total Synthesis of Colchicine via the Scott's Intermediate. Palladium (0) - Mediated meta-Ethynylation of a Tropolone Derivative

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Colchicine was formally synthesized from 4-hydroxytropolone in a few-step sequence. The key step of the synthesis is palladium(0)-catalyzed ethynylation of 4-(trifluoromesyloxy)-2-methoxytropone. The present method has a potential in synthesizing various structurally modified colchicine derivatives.

Up to the present, numerous syntheses of colchicine (1) have been reported.^{1,2)} Since the unique physiological activity of 1 has been a focus of interest, development of an effective synthesis is desirable. Herein, we report a short formal synthesis of 1 based on the troponoid chemistry.

Among the recently developed transition metal-mediated C-C bond formation reactions, there might be some promising method applicable to troponoid chemistry. A notable development was reported by Stille,³⁾ i.e., the palladium metal-catalyzed condensation of carbon monoxide and 1-(tributylstannyl)-2-(trimethylsilyl) ethene with 2-(trifluoromethylsulfonyloxy) tropone (**A**) afforded 2-(3-trimethylsilylpropenoyl) tropone (**B**) in a good yield (Scheme 1). In addition, diphenylethyne formation by the palladium catalyzed oxidative

$$O$$
-SO₂CF₃
 O + O

coupling of iodobenzene and ethyne, developed by Sonogashira et al.,⁴⁾ has another superior feature in view of potential in troponoid chemistry, i.e., this ethyne substitution occurred smoothly with an aryl halide and a terminal acetylenic hydrocarbon. We have now applied this palladium(0)-mediated ethynylation to the troponoid chemistry.

For a sake of convenience, 2,5-di(trifluoromethylsulfonyloxy)tropone (2, 2,5-di(trifluoromesyloxy)tropone),⁵⁾ which could be prepared from 5-hydroxytropolone, was at first subjected for the reaction; the reaction of 2 with 1-hexyne in the presence of tetrakis-(triphenylphosphine)palladium(0) complex generated from palladium(II) chloride, triphenylphosphine, copper(I) iodide, benzyltrimethylammouium chloride, and aqueous sodium hydroxide under argon atmosphere, serves itself as the reactive agent to afford 2-(trifluoromesyloxy)-5-(2-phenylethynyl)tropone (3) in 61% yield. However, the transformation of the 2-(trifluoromesyloxy) group was difficult.

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To overcome the difficulty, the trifluoromesyloxy group on the C-2 position should be changed to an electron-donating alkoxyl group, and 2-methoxy-5-(trifluoromesyloxy)-tropone (4) was studied; under the similar conditions, but triethylamine instead of sodium hydroxide, 4 and 1-heptadecyne gave 5-(1-heptadecynyl)-2-methoxytropone (5) in 86% yield. This result with the higher homologue of 1-alkyne is quite promising. Furthermore, the reaction of 4 with 3-(3,4,5-trimethoxyphenyl)-1-propyne (6) gave 2-methoxy-5-[3-(3,4,5-trimethoxyphenyl)-1-propynyl]tropone (7) (Scheme 2).

These findings on the preliminary studies led to extend the reaction with a metasubstituted derivative, 2-methoxy-6-(trifluoromesyloxy) tropone as the direct starting material leading to 1. Although, in general, direct introductions of carbon substituents on the meta position of tropolones are limited, 6) the required 6-(trifluoromesyloxy)-2-methoxytropone (8) was prepared as follows: heating of 4-chlorotropolone (9) in a sealed tube with a mixture of acetic anhydride and trifluoroacetic acid⁷⁾ at 80 °C for 35 h gave 2,6-diacetoxytropone (10), which was then converted into 6-acetoxy-2-methoxytropone (11) by mild acetic acid treatment followed by methylation with diazomethane. A mild hydrolysis of 11 with aqueous acetic acid gave 6-hydroxy-2-methoxytropone (12). Then, 12 was treated with trifluoromesyl chloride to give 6-(trifluoromesyloxy)-2-methoxytropone (8). The palladium(0)mediated coupling, as above, of 8 with 3-(3,4,5-trimethoxyphenyl)-1-propyne (6) in N, Ndimethylformamide with added dimethylamine afforded 2-methoxy-6-[3-(3,4,5-trimethoxyphenyl)-l-propynyl]tropone (13). Catalytic hydrogenation of 13 gave 2-methoxy-6-[3-(3,4,5-trimethoxyphenyl) propyl]tropone (14), which was further hydrolyzed to 4-[3-(3,4,5trimethoxyphenyl) propyl]tropolone (15). The ¹H NMR spectra of 14 and 15 were identical with those of the authentic samples (Scheme 3).

This led to an efficient formal synthesis of colchicine by obtaining desacetoamido-B-seco-colchiceine (15), the intermediate in the former total synthesis of 1.

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- 5) The following ¹H and/or ¹³C NMR data were compiled for characterization of key compounds: **2**: $\sigma(H) = 7.02$ (1H, ddd, J = 10.3, 2.7, 0.7 Hz), 7.23 (1H, dd, J = 13.2, 2.7 Hz), 7.35 (1H, d, J = 10.3 Hz), and 7.41 (1H, d, J = 13.2, 0.7 Hz); $\sigma(C) = 111.6$ (q, $J_{C-F} = 320.8$ Hz), 117.3, 120.9, 122.1, 125.5, 125.6, 132.7, 141.6, 152.8, 155.7, and 176.6.
 - 3: δ (H) =0.95(3H, t, J =7.3 Hz), 1.39-1.78 (4H, m), 2.45 (2H, t, J =7.0 Hz), 7.08 (1H, d, J =10.3 Hz), 7.23 (1H, d, J =12.8 Hz), 7.25 (1H, dd, J =10.3, 1 Hz), and 7.33 (1H, dd, J =12.8, 1.5 Hz); δ (C) =13.5, 19.4, 22.0, 30.3, 82.3, 100.2, 116.3, 121.0, 127.7, 132.1, 133.3, and 177.6.
 - 4: δ (H) = 4.03 (3H, s), 6.62 (1H, d, J = 11 Hz), 7.06 (1H, ddd, J = 11, 2.5, 0.5 Hz), 7.16 (1H, dd, J = 13, 2.5 Hz), and 7.23 (1H, dd, J = 13, 0.5 Hz).
 - 5: $\delta(C) = 14.0$, 23.6, 24.7, 28.8, 29.0, 29.2, 29.4, 29.50 (3C), 29.53 (3C), 31.8, 56.2, 82.1, 93.1, 111.9, 124.4, 135.1, 135.9, 139.3, 164.5, and 179.7.

- 6: δ (H) = 2.21 (1H, t, J = 2.5 Hz), 3.56 (1H, d, J = 2.5 Hz), 3.83 (3H, s), 3.87 (6H, s), and 6.58(2H, s).
- 7: δ (H) = 3.77 (2H, s), 3.84 (3H, s), 3.88 (6H, s), 3.96 (3H, s), 6.59 (2H, s), 6.65 (1H, dd, J = 10, 1.5 Hz), 7.14 (1H, d, J = 12 Hz), 7.26 (1H, dd, J = 10, 1.5 Hz), 7.30 (1H, d, J = 12 Hz); δ (C) = 25.9, 56.2 (2C), 56.3, 60.7, 84.0, 89.9, 104.9 (2C), 111.8, 123.7, 128.3, 131.8, 135.6, 136.0, 139.0, 153.2 (2C), 164.8, and 179.7.
- 8: δ (H) = 4.00 (3H, s), 6.78-6.85 (2H, m), 7.17 (1H, d, J = 2.5 Hz), 7.22 (1H, dd, J = 10.5, 1 Hz); δ (H) = 56.5, 111.3, 118.3 (q, J = 321 Hz), 122.6, 128.1, 133.4, 156.2, 166.2, and 176.7.
- **9**: $\delta(H) = 7.11 7.24$ (3H, m) and 7.50 (1H, d, J = 2 Hz).
- **10**: δ (H) = 2.31 (3H, s), 2.35 (3H, s), 6.84 (1H, ddd, J = 10.5, 2.5, 1 Hz), 7.01 (1H, d, J = 2.5 Hz), 7.06 (1H, t, J = 10.5 Hz), and 7.13 (1H, dd, J = 10.5, 1 Hz).
- **13**: δ (H) = 3.77 (2H, s), 3.84 (3H, s), 3.88 (6H, s), 3.93 (3H, s), 6.58 (2H, s), 6.63 (1H, d, J = 8.5 Hz), 6.69-6.87 (2H, m), and 7.42 (1H, s); δ (C) = 26.0, 56.1 (2C), 56.3, 60.8, 84.1, 92.5, 104.9 (2C), 111.2, 130.7, 131.2, 131.3, 133.1, 136.8, 139.5, 153.3 (2C), 165.7, and 179.0.
- **14**: δ (H) =1.89-2.01(2H, m), 2.54-2.63(2H, m), 3.83(3H, s), 3.85(6H,s), 3.93(3H, s), 6.38(2H, s), 6.64(1H, d, J =10 Hz), 6.74(1H, d, J =11 Hz), 6.98(1H, m), and 7.18(1H, s); δ (C) =32.2, 35.5, 40.1, 56.0(2C), 58.3, 60.8, 105.1(2C), 111.5, 130.6, 131.5, 136.1, 136.4, 137.2, 151.7, 153.1(2C), 164.9, and 179.6.
- **15**: δ (H) =1.93-2.04 (2H, m), 2.63 (2H, t, J =7.5 Hz), 2.67 (2H, t, J =7.5 Hz), 3.83 (3H, s), 3.86 (6H, s), 6.39 (2H, s), 6.92 (2H, d, J =9 Hz), and 7.20-7.33 (3H, m); δ (C) =32.8, 35.6, 56.0 (2C), 60.8, 105.2 (2C), 122.2, 125.2, 129.4, 136.2 (2C), 137.0, 137.1, 153.1 (2C), 153.9, and 170.9.
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