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Insecticidal Activity of Lignan Analogs Chromano-Sesquilignans with a Dihydroprecocene Skeleton+

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Two stereoisomers of a chromano-lignan were synthesized as the analogs of an insecticidal 1,4-benzodioxano-sesquilignan by employing the well-known anti-juvenile hormone, precocene I, as a building block for the synthesis. Both isomers, (\pm) - $(1S^*,2R^*,$ $5R^*,6S^*$) - 2 -(2,6 - dimethoxyphenoxy) - 6 -(2,2 - dimethyl 7 - methoxy - 3,4 - dihydro - 2H chromen-6-yl)-1-hydroxy-3,7-dioxabicyclo[3.3.0] octane and its 6R* isomer, showed neither insect growth-regulatory property nor insecticidal activity.

INTRODUCTION

Haedoxan A, a unique sesquilignan belonging to the class of neolignan of 1,4benzodioxane type (Fig. 1), was isolated as the insecticidal constituent of Phryma leptostachya L. (Taniguchi et al., 1989). By a sustaining investigation on the correlation between structure and activity of the lignan, it has been made evident that the 6 - methoxy- 2 - methoxymethyl - 3 - (3,4 - methylenedioxyphenyl) - 1,4 - benzodioxan - 7 - yl moiety is indispensable for the potent activity of haedoxan (Yamauchi et al., 1992; Yamauchi and Taniguchi, 1992a,b).

In a scope of the structure-activity relationship of the bioactive lignan, it was proposed to estimate the activity of two chromano-analogs such as CA1 and CA2 (Fig. 1) replacing the 1"-or 4"-oxygen atom of the haedoxan molecule by a methylene group. On purpose to establish a synthetic route for these analogs, the synthesis of (\pm) - $(IS^*,$ $2R^*$, $5R^*$, $6S^*$) - 2 - (2,6 - dimethoxyphenoxy) - 6 - (2,2 - dimethyl - 7 - methoxy - 3,4 - dihydro-2H-chromen-6-yl)-1-hydroxy-3,7-dioxabicyclo[3.3.0] octane (15a) has been carried out as a preliminary to the application of the method developed for the synthesis of haedoxans and their analogs (Ishibashi and Taniguchi, 1989; Yamauchi et al., 1990). In the synthesis, a stereoisomer (15b) with $6R^*-(2,2-\text{dimethyl-}7-\text{methoxy-}3,4-\text{dihydro})$ -2H-chromen-6-yl) group was also obtained as a by-product. Bioactivity of these chromano-analogs was assessed on the housefly and silkworm.

MATERIALS AND METHODS

The ¹H-NMR All melting points (mp) and boiling points (bp) were uncorrected.

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Fig. 1. Structures of haedoxan A and chromano-lignans.

spectra were determined with JEOL FX100 spectrometer in deuterochloroform with tetramethylsilane ($\delta_{\rm H}$ 0.0) as an internal standard. Gravity chromatography and medium-pressure column chromatography were performed with Wakogel C-300. Preparative TLC was conducted with $60F_{254}$ pre-coated silica gel sheets in thickness of 0.5mm supplied by E. Merck.

2.2-Dimethyl-7-methoxychroman (1). Precocene I (2,2-dimethyl-7-methoxychromene, Aldrich Chemical, 0.88g,4.63mmol) and a palladium catalyst (Nacalai Tesque, 260mg, a 5% Pd on carbon) in ethyl acetate (12ml) was vigorously stirred for 1.5hr at room temperature under a hydrogen atmosphere (1 atm). The reaction mixture was filtered, and the filtrate was concentrated. The residue was purified by column chromatography (9:1 hexane-ethyl acetate) to give a colorless oil (0.73g, 82%). **Anal.** Found: C, 75.21; H, 8.47. Calcd. for $C_{12}H_{16}O_2$: C, 74.97; H, 8.39%. 100 MHz NMR $\delta_H(CDCl_3)$: 1.24 (6H, s), 1.68 (2H, t, J=6), 2.62 (2H, t, J=6), 3.66(3H, s), 6.14-6.46 (2H, m), 6.70-6.98 (1H, m).

2,2-Dimethyl-6-formyl-7-methoxychroman (2). Chroman 1 (0.73g, 3.80mmol) and phosphorous oxychloride (5.70mmol) in dimethyl formamide (2.6ml) was heated at 60°C for lhr. After the reaction mixture had been treated with 10% aqueous sodium carbonate (50ml), it was extracted with ethyl acetate. The extract was dried and concentrated, and the residue was purified by column chromatography to give 2 (0.80g, 96%) as colorless crystals, mp 74-76°C (isoPr₂O). **Anal.** Found: C, 71.02; H, 7.34. Calcd. for $C_{13}H_{16}O_3$: C, 70.89; H, 7.32%. 100 MHz NMR δ_H (CDCl₃): 1.38 (6H, s), 1.82 (2H, t, J=6), 2.76 (2H, t, J=6), 3.88 (3H, s), 6.18 (1H, s), 7.62 (1H, s), 10.28 (1H, s).

4-Ethenyl-2-[(trimethylsilyl)oxy]-4,5-dihydrofuran (3) was prepared from 3-vinyl-butanolide according to the method described in the literature (Ainsworth *et al.*, 1972) to obtain a colorless oil, bp 84-87°C/33mmHg. **Anal.** Found: C, 59.04; H, 8.94. Calcd. for C₉H₁₆O₂Si: C, 58.65; H, 8.75%. 100 MHz NMR $\delta_{\rm H}$ (CDCl₃): 0.28 (9H, s), 3.44 -3.68 (1H, m), 3.98 (1H, dd, J = 8,6), 4.43 (1H, dd, J = 8,8), 4.87-5.38 (2H, m), 5.47 (1H, broad d, J = 6), 5.63-6.02 (1H, m).

 (\pm) - $(3S^*,4R^*,1'R^*/S^*)$ -3-[1'-Hydroxy-1'-(2,2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl)methyl]-4-vinyldihydro-2(3H)-furanones (5). A mixture of trityl chloride (Nacalai Tesque, 45mg,0.16mmol) and tin (II) chloride (Aldrich Chemi-

cal, 30.3mg, 0.16mmol) in dry dichloromethane (2ml) was stirred for 30min at room temperature and cooled to -75°C. To the cooled and stirred mixture a solution of 2 (0.71g, 3.23mmol) and 3 (0.89g, 4.84mmol) in dichloromethane (6ml) was added. the mixture had been stirred for 6hr at the same temperature, it was treated with 5% aqueous sodium bicarbonate and extracted with ethyl acetate. The extract was passed through a silica gel column (3:1 hexane-ethyl acetate) to obtain trimethylsilyl ether 4 (1.04g, 80%). The silvl ether was readily reacted with tetra-n-butylammonium fluoride (Janssen Chemica, 3.1ml, a 1 M solution in tetrahydrofuran, 3.1mmol) in dry tetrahydrofuran (9ml) at 0°C for 2hr. The reaction mixture was treated with 5% aqueous ammonium chloride (2ml) and extracted with ethyl acetate. The extract was purified by chromatography (3:1 hexane-ethyl acetate) to obtain 5 (0.56g, 66%) as a colorless oil. **Anal.** Found: C, 68.97; H, 7.49. Calcd. for C₁₉H₂₄O₅: C, 68.65; H, 7. 100 MHz NMR $\delta_{H}(CDCl_3)$: 1.26 (6H, s), 1.70 (2H, t, J=6), 2.62-3.30 (2H, m), 2.64 (2H, t, J=6), 3.66 (3H, S), 3.81 (1H, dd, J=10.8), 4.00 (1H, d, J=4), 4.28 (1H, dd, J=10.8)8), 4.50-4.82 (2H, m), 4.98-5.30 (1H, m), 5.34-5.50 (1H, m), 6.22 (1H, s), 7.03 (0.4H, s), 7. 08(0.6H, s).

The following compounds were obtained by the same reactions as those described in the literatures (Ishibashi and Taniguchi, 1986; Yamauchi and Taniguchi, 1992 etc.), the data being briefly described here.

(±)-(3S*,4R*,1'R*/S*)-3-{1'-[(tert-Butyldimethylsilyl)oxy]-1'-(2,2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl)methyl}-4-vinyldihydro-2(3H)-furanones (6). Silylation [5 1.66mmol (0.55g), tert-butyldimethylsilyl triflate 2.48mmol, 2,6-lutidine 3.35mmol in dichloromethane 4ml at room temperature for 4hr]. A colorless oil (0.55g, 74%). **Anal.** Found: C, 67.01; H, 8.54. Calcd. for $C_{25}H_{38}O_5Si$: C, 67.23; H, 8.58%. 100 MHz NMR δ_H(CDCl₃): -0.13 (1.2H, s), -0.08 (1.8H, s), 0.08 (1.2H, s), 0.09 (1.8H, s), 0.94 (9H, s), 1.30 (2.4H, s), 1.34 (3.6H, s), 1.78 (2H, t, J=6), 2.54-3.42 (2H, m), 2.68 (2H, t, J=6), 3.72 (1.2H, s), 3.75 (1.8H, s), 3.92 (1H, dd, J=10, 8), 4.40 (1H, dd, J=10, 8), 4.58-4.78 (1H, m), 4.98-5.46 (2H, m), 5.46-5.78, (1H, m), 6.06 (1H, s), 7.10 (0.6H, s), 7.26 (0.4H, s).

(±)-(2 R^* ,3 R^* ,1' R^* /S*)-2-{1'-[(tert-Butyldimethylsilyl)oxy]-1'-(2,2-dimethyl-7 - **methoxy** -3,4-**dihydro** -2H-chromen-**6**-yl)methyl}-3-**vinyl**-1,4-butanediols (7). Reduction [**6**1.21mmol (0.54g), lithium aluminium hydride 1.58mmol in ethyl ethertetrahydrofuran (2:1) 18ml at -10°C for 1.5hr]. A colorless oil (0.32g, 58%). **Anal.** Found: C, 66.99; H, 9.56. Calcd. for C₂₅H₄₂O₅Si: C, 66.63; H, 9.39%. 100 MHz NMR δ_H(CDCl₃): -0.30 (3H, s), 0.00 (3H, s), 0.86 (9H, s), 1.34 (6H, s), 1.60 (1H, broad s), 1.68-1.98 (1H, m), 1.78 (2H, t, J=6), 2.70 (2H, t, J=6), 2.72-3.04 (1H, m), 3.24-4.02 (5H, m), 3. 81 (3H, s), 5.02-5.40 (3H, m), 5.76-6.20 (1H, m), 6.34 (1H, s), 7.12 (1H, broad s).

(±)- $(2R^*/S^*,3R^*,4R^*,1'R^*/S^*)$ - 4 - {1'-[(tert - Butyldimethylsilyl)oxy] - 1'-(2,2 - dimethyl-7 - methoxy - 3,4 - dihydro -2H - chromen-6-yl)methyl}-2-hydroxy -3-hydroxymethyltetrahydrofurans (8). Oxidation [7 0.69mmol (0.31g), a catalytic amount of osmium tetroxide, N-methylmorpholine N-oxide 0.77mmol, tert-butyl alcohol 5ml, water 5ml, acetone 10ml at room temperature for 20 hr under nitrogen in the dark]. Oxidation [sodium metaperiodate 0.84mmol in ethyl acetate 50ml at room temperature for 40hr]. A viscous oil (0.28g, 90%). **Anal.** Found: C, 63.96; H, 8.99. Calcd. for C_{24} H₄₀O₆Si: C, 63.68; H, 8.91%. 100 MHz NMR δ_{H} (CDCl₃): -0.36 (3H, s), -0.06 (3H, s), 0. 86 (9H, s), 1.34 (6H, s), 1.72 (1H, broad s), 1.78 (2H, t, J=6), 2.30-2.54 (1H, m), 2.70 (2H,

- t, J=6), 2.86-3.30 (2H, m), 3.42-4.45 (4H, m), 3.68-3.82 (3H, s x 2), 5.14 (1H, d, J=6), 5. 48 (1H, d, J=2), 6.28-6.30 (1H, s x 2), 7.10 (1H, broad s).
- (\pm)-($3R^*$, $4R^*$, $1'R^*/S^*$)-4-{1'-[(tert-Butyldimethylsilyl)oxy]-1'-(2,2-dimethyl-7-methoxy -3,4-dihydro -2H chromen 6 yl) methyl} 3 hydroxymethyldihydro 2(3H)-furanones (9). Oxidation [8,0.60mmol (0.27g), silver carbonate 0.62mmol on Celite in refluxing benzene 12ml for 10min]. A colorless oil (0.21g, 78%). Anal. Found: C, 63.65; H, 8.42. Calcd. for $C_{24}H_{38}O_6Si$: C, 63.97; H, 8.50%. 100 MHz NMR $\delta_H(CDCl_3)$: -0.30 (3H, s), 0.08 (3H, s), 0.90 (9H, s), 1.34 (6H, s), 1.80 (2H, t, J=6), 2.70 (2H, t, J=6), 2.74-2.98 (2H, m), 3.02-3.30 (1H, m), 3.78 (3H, s), 3.90-4.26 (4H, m), 5.21 (1H, d, J=6), 6. 32 (1H, s), 7.06 (1H, broad s).
- (±)-(4 R^* ,1' R^* /S*)- 4 -{ 1'- [(tert-Butyldimethylsilyl)oxy]-1'-(2,2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-ylmethyl}-3-methylenedihydro-2(3H)- furanones (10). Mesylation [9 0.44mmol, (0.20g), mesyl chloride 0.67mmol, triethylamine 0.88mmol in dry benzene 10ml at room temperature for 2hr.] Desulfonation [1,8-diazabicyclo [5.4.0] undec-7-ene 0.669mmol, 30min, room temperature]. The reactions were carried out as a one-pot reaction. A colorless oil (0.17g, 89%). Anal. Found: C, 66.89; H, 9.44. Calcd. for $C_{24}H_{36}O_5Si$: C, 66.63; H, 9.39%. 100 MHz NMR δ_H (CDCl₃): -0.22 (3H, s), 0.02 (3H, s), 0.88 (9H, s), 1.36 (6H, s), 1.80 (2H, t, J=6), 2.68 (2H, t, J=6), 3.12-3.36 (1H, m), 3.76 (3H, s), 4.12-4.36 (2H, m), 5.09 (1H, d, J=6), 5.24 (1H, broad s), 6.26 (1H, broad s), 6.32 (1H, s), 7.02 (1H, broad s).
- (±)-(3S*,4R*, 1'R*/S*)-4-{1'-[(tert-Butyldimethylsilyl)oxy]-1'-(2,2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl)-1-hydroxymethyl}-3-hydroxy-3-hydroxymethyldihydro-2(3H)-furanones (11). Oxidation [10 0.37mmol (0.16g), a catalytic amount of osmium tetroxide, N-metylmorpholine N-oxide 0.41mmol, water 5ml, tert-butyl alcohol 5ml, acetone 10ml for 14hr at room temperature under nitrogen in the dark]. A colorless oil (0.17g, 81%). Anal. Found: C, 62.20; H, 8.34. Calcd. for C_{24} H₃₈O₇: C, 61.77; H, 8.21%. 100 MHz NMR $\delta_{\rm H}$ (CDCl₃): -0.32 (3H, s), 0.06 (3H, s), 0.90 (9H, s), 1.34 (6H, s), 1.66 (1H, s), 1.80 (2H, t, J=6), 2.70 (2H, t, J=6), 2.94-3.26 (1H, m), 3.10 (1H, s), 3.78 (3H, s), 3.82-4.22 (4H, m), 5.29 (1H, d, J=6), 6.32 (1H, s), 7.06 (1H, s).
- (±)-(3S*,4R*,1'R*/S*)-4-[1'-Hydroxy-1'-(2,2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl)methyl]-3-hydroxy-3-hydroxymethyldihydro-2(3H)-furanones (12). Desilylation [11 0.279mmol (0.13g)), tetra-n-butylammonium fluoride 0.335mmol in dry tetrahydrofuran 5ml at 0°C for lhr]. A colorless oil (61.7mg 63%). Anal. Found: C, 61.79; H, 7.09. Calcd. for $C_{18}H_{24}O_7Si$: C, 61.35; H, 6.87%. 100 MHz NMR $\delta_H(CDCl_3)$: 1.34 (6H, s), 1.78 (2H, t, J=6), 2.68 (2H, t, J=6), 3.02-3.38 (1H, m), 3.38-3.66 (2H, broad), 3.78 (3H, s), 3.82-4.18 (5H, m), 5.11 (1H, dd, J=8,4), 6.36 (1H, s), 7.02(1H, s).
- (\pm)-($1S^*,5R^*,6R^*/S^*$)-6-[2,2-Dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl)-1-hydroxy-3,7-dioxabicyclo [3.3.0] octan-2-ones (13). To a stirred and ice-cooled solution of 12 (57.5mg, 0.163mmol) in dry dichloromethane (5ml) containing molecular sieves 4A was added a catalytic amount of 10-camphorsulfonic acid. After the stirring had been continued for 1.5hr at 0° C, the reaction mixture was treated with a few drops of triethylamine and concentrated. The residue was purified by preparative TLC (Silica gel 60F $_{254}$ pre-coated plate of 0.5mm thickness/hexane-AcOEt 1:1) to give colorless crystals (43.6mg, 80%), mp 151- 153° C. **Anal.** Found: C, 64.89; H, 6.72. Calcd. for $C_{18}H_{22}O_6$: C, 64.66; H, 6.63%.

100 MHz NMR $\delta_{\rm H}({\rm CDCl_3})$: 1.34 (6H, s, 2'-(C H_3)₂), 1.78 (2H, t, J=6,3'-H), 2.72 (2H, t, J=6', 4-H), 3.13 (0.5H, m, 5 α -H), 3.49 (0.5H, m, 5 α -H), 3.63-4.76 (5H, m, 1-OH, 4-C H_2 , 8-H,), 3.76 (3H, s, OC H_3), 5.10 (0.5H, d, J =6,6 β -H), 5.34 (0.5H, d, J =6,6 α -H), 6.32 (0.5H, s, 8'-H), 6.36 (0.5H, s, 8'-H), 7.12 (0.5H, s, 5'-H), 7.18 (0.5H, s, 5'-H).

(±)-(1S*,2R*/S*,5R*,6R*/S*)-1,2-Dihydroxy-6-(2,2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl)-3,7-dioxabicyclo[3.3.0] octanes (14). Reduction [13 0.112mmol (37.4mg), diisobutylaluminium hydride 0.425mmol in toluene 10ml for 1.5hr at -70°C under nitrogen] . A colorless resinous material (30.1mg, 80%). **Anal.** Found: C, 64.57; H, 7.20. Calcd. for $C_{18}H_{24}O_6$: C, 64.27; H, 7.19%. 100 MHz NMR δ_H (CDCl₃): 1.34 (6H, s), 1.76 (2H, t, J=6), 2.56 (0.5H, m), 2.72 (2H, t, J=6), 3.24 (0.5H, m), 3.32 (1H, broad s), 3.74-3.78 (3H, s×2) 3.60-4.40 (5H, m), 4.83 (1H, d, J=6), 5.12 (1H, broad s), 6.32-6.36 (1H, s×2), 7.18-7.26 (1H, s×2).

(\pm)-($1S^*$, $2R^*$, $5R^*$, $6R^*/S^*$)-2-(2,6-Dimethoxyphenoxy)-6-(2,2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl)-1-hydroxy-3,7-dioxabicyclo[3.3.0] octanes (15a and 15b). Acetalization[14 0.081mmol(27.3mg), 2,6-dimethoxyphenol 2.01mmol, a catalytic amount of pyridinium P-toluenesulfonate in benzene 2ml, reflux for 1.5hr under nitrogen with azeotropic removal of the water]. Preparative TLC[$60F_{254}$ precoated silica gel plate of thickness in 0.5mm/5% ethyl alcohol in a mixture of hexane and ethyl acetate (2:1)].

 $6S^*(\alpha)$ -isomer 15a. A colorless resinous material (14.5mg, 38%). Rf 0.55. **Anal.** Found: C, 66.28; H, 6.91. Calcd. for C₂₆H₃₂O₈: C, 66.09; H, 6.83%. 100 MHz NMR δ_H (CDCl₃): 1.34 (6H, s, 2"-(CH₃)₂), 1.78 (2H, t, J=6,3"-H₂), 2.54-2.82 (1H, m, 5β-H), **2.74** (2H, t, J=6,4"-H₂), 3.71 (1H, d, J=10,8α-H), 3.76 (3H, s, 7"-OCH₃), 3.88 (6H, s 2'-OCH₃, 6'-OCH₃) 4.10 (1H, dd, J=9, 2, 4α, -H), 4.28 (1H, broad s, 1-OH), 4.33 (1H, d, J=10,8α-H), 4.60 (1H, dd, J=9, 7, 4β-H), 4.90 (1H, d, J=6,6α-H), 5.38 (1H, s, 2α-H), 6.34 (1H, s, 8"-H), 6.61 (1H, d, J=9,3'/5'-H), 6.62 (1H, d, J=7,3'/5'-H), 7.07 (1H, m, 4'-H), 7.33 (1H, s, 5"-H).

 $6R^*(β)$ -isomer **15b.** A colorless resinous material (llmg, 28%). Rf 0.51. **Anal.** Found: C, 66.25; H, 6.88. Calcd. for C₂₆H₃₂O₈: C, 66.09; H, 6.83%. 100 MHz NMR $δ_H(CDCl_3)$: 1.33 (3H, s, 2"-C H_3), 1.34 (3H, s, 2"-C H_3), 1.78 (2H, t, J=6, 3"- H_2), 2.70 (2H, t, J=6, 4"- H_2), 3.13 (1H, m, 5β-H), **3.50-4.30** (4H, m), **3.76** (3H, s 7"-OC H_3), 3.88 (6H, s, 2"-OC H_3 , 6"-OC H_3), 4.42 (1H, s, 1-OH), 5.36 (1H, s, 2α-H), 5.38 (1H, d, J=4,6β-H), 6.32 (1H,s, 8"-H), 6.59 (1H, d, J=10,3"/5"-H), 6.60 (1H, d, J=8,3"/5"-H), **7.04** (1H, m, 4"-H), 7.20 (1H, s, 5"-H).

RESULTS AND DISCUSSION

The synthesis of the chromano-lignans is shown in Scheme. Precocene I, 7-methoxy-2,2-dimethylchromene, was readily converted to 7-methoxy-2,2-dimethyl-6-formylchroman 2 by sequential hydrogenation and the Vilsmeier formylation in a 79 % overall yield. The chromancarbaldehyde scarcely reacted with the lithium enolate of (\pm) -3-vinylbutanolide (Kondo and Mori, 1974) in dry tetrahydrofuran; this result was dissimilar to those of the aldol condensation for an alkoxybenzaldehyde and 1,4-benzodioxanecarbaldehydes using in the earlier syntheses of phrymarolin (Ishibashi and Taniguchi, 1986, 1989) and haedoxans (Yamauchi *et al.*, 1992; Yamauchi and Taniguchi, 1992a, b). A cross aldol reaction using a cerium enolate (Imamoto *et al.*,

1983) of the butanolide was also not progressive. Finally, in the application of a modified aldol reaction using Lewis acid (Mukaiyama et al., 1987), a trimethylsilyl enol ether 3 preparative from (\pm) -3-vinylbutanolide by the reported method (Ainsworth et al., 1972) was reacted with chromancarbaldehyde 2 in the presence of trityl chloride and tin (II) chloride to produce trimethylsilylated (±)-3,4-trans-aldol 4 as a mixture of the erythro and threo adducts in an 80% yield. In the present analog synthesis, every reaction for the intermediates 4-12 was successively carried out by employing a mixture of erythro and threo isomers, since each isomer of the products of those reactions was unable to be isolated in pure form by chromatographic separation. Thus, the mixture of trimethylsilyl ethers 4 was desilylated by treating with tetra-n -butylammonium fluoride in tetrahydrofuran (Corey et al., 1981) to afford (\pm) -3,4trans-aldols 5 in a 53% overall yield from 2, the NMR data of 5 suggesting the ratio of the erythro/threo isomers to be 3/2. A tert-butyldimethylsilylation (Corey and Venkateswarlu, 1972; Stewart and Miller, 1980) of the aldols was followed by sequential reduction, oxidation (3 steps), mesylation, elimination, glycolization, desilylation, and acid-catalyzing intramolecular cyclization accompaying dehydration (Ishibashi and Taniguchi, 1986, 1989; Yamauchi et al., 1992) to give rise diol 7, lactol 8, α hydroxymethyl-y-butyrolactone 9, α -methylene- γ -butyrolactone 10, α -hydroxy- α hydroxymethyl-y-butyrolactone 11, triol 12, and the key intermediary 1-hydroxy-6-(2, 2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl)-3,7-dioxabicyclo[3.3.0]octan-2-one (13), respectively. The key intermediate 13 was obtained as a mixture of equimolecular amounts of $1S^*, 5R^*, 6S^*$ and $1S^*, 5R^*, 6R^*$ isomers in a 12% overall yield for ten steps from aldols 5. In a NMR spectrum of 13, the multiplets at 3.13ppm (0.5H) and 3.49ppm (0.5H) were assigned to each 5α -proton of the $6S^*(6\alpha)$ and $6R^*(6\beta)$ isomers, respectively.

The $6S^*(6\alpha)$ isomer would be thermodynamically more stable than the $6R^*(6\beta)$ isomer and expected to be preferentially formed from triol 12 by the acid-catalyzing cyclyzation reaction. Really, a triol bearing an alkoxyphenyl or a 1,4-benzodioxanyl group in place of the chromanyl group of 12 had selectively produced a 6α isomer (Fig. 2) under a similar reaction condition as that used for the cyclyzation of 12 (Ishibashi and Taniguchi, 1988, 1989; Yamauchi et al., 1992; Yamauchi and Taniguchi, 1992a, b).

Fig. 2. Stereoselective cyclization of the triols having a polyalkoxyphenyl group,

Scheme Synthesis of dihydroprecocene-lignan

1) OsO₄, NMO m) nBu_4F n) CSA, MS(4A) o) DIBAL-H p) 2,6-dimthoxyphenol, cat. PPTS q) chromatographic separation

Scheme (continued)

The non-stereoselective cyclization of the trio1 12 remained to be elucidated.

The $6\alpha/\beta$ - dihydrochromenyl -3,7 - dioxabicyclooctanones 13 were together reduced with diisobutylaluminium hydride, and the resulting lactols 14 were reacted with 2,6-dimethoxyphenol in the presence of pyridinium p-toluenesulfonate (Takano et al., 1988) to selectively produce 2α -acetal as the equimolecular mixture of (\pm) - $(1S^*, 2R^*, 5R^*, 6S^*)$ -2-(2,6-dimethoxyphenoxy)-1-hydroxy-6-[2,2-dimethyl-7-methoxy-3,4-dihydro-2H-chromen-6-yl]-3,7-dioxabicyclo-[3.3.0] octane (15a) and its (\pm) - $(1S^*, 2R^*, 5R^*, 6R^*)$ -isomer (15b). These diastereomers were purified by preparative TLC to obtain in 1.73% and 1.27% overall yields from 2, respectively. Their stereochemistry was unequivocally determined on the basis of the chemical shift value of 5α -H (Taniguchi et al., 1989; Pelter et al., 1976); the protons of the $6S^*(\alpha)$ and $6R^*(\beta)$ isomers displayed a multiplet at 2.68ppm and 3.13ppm in their NMR spectra, respectively.

The compounds 15a and 15b, which involved in the molecules two structures of both anti-juvenile hormone (Bowers *et al.*, 1976) and insecticidal haedoxan, were totally inactive to the 4-day old housefly, *Musca domestica* (SRS), at a dose level of $10\mu g$, and also ineffective on the growth of the 4th instar larvae of the silkworm, *Bombyx mori* (C. 146 x N. 137), at $200\mu g$ per larva, the housefly and silkworm being

treated with each compound in a $1\mu l$ of acetone solution at the dorsal thorax and the head-thorax by a micro-applicator, respectively.

The chromano-lignan CA2 with a 7-methoxy-3-methoxymethyl-2-(3,4-methylenedioxyphenyl)-3,4-dihydro-2H-chromen-6-yl group has been recently synthesized by applying the method accomplished in this preliminary synthesis, and the compound was proved to be insecticidally active to the housefly (Gotanda $et\ al.,\ 1993$). Combining the assay results of **15a** and CA2, it is implied that the 2"-(3,4-methylenedioxyphenyl) and 3"-methoxymethyl groups may significantly contribute to the insecticidal activity of chromano-analogs of haedoxan.

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