## Catalytic Asymmetric Hydrolysis: Asymmetric Hydrolytic Protonation of Enol Esters Catalyzed by Phase-Transfer Catalysts

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## Asymmetric Ester Hydrolysis: Catalytic Asymmetric Protonation of Enolesters Catalyzed by Phase Transfer Catalysts

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#### 1. General.

Unless otherwise stated, all the reactions were performed in flame-dried glassware under a nitrogen atmosphere using dry solvents. Commercial reagents were used as received. <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded on a JEOL spectrometer ECS-400, ECS-600 and AL-400. Data for <sup>1</sup>H NMR spectra are reported as follows: chemical shift ( $\delta$  ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dd = doublet of doublets, ap = apparent, sex =sextet, sep = septet), integration, coupling constant (Hz) and assignment. The enantiomeric excesses were determined by GC or HPLC analysis employing a chiral stationary phase column specified in the individual experiment, by comparing the samples with the appropriate racemic mixtures. GC analysis was carried out using Agilent GC 6850 series II equipped with InertCap CHIRAMIX Column (length 30 m, i.D. 0.25 mm, df. 0.25 µm) from GL Sciences Inc. and CHIRASIL-DEX CB (length 25 m, i.D. 0.25 mm, df. 0.25 µm) from Varian using helium as a carrier gas. GC yields were determined by employing HP-1 (length 30 m, i.D. 0.320 mm, df. 0.25 µm) or HP-INNOWAX (length 30 m, i.D. 0.320 mm, df. 0.25 µm) column from Agilent Technologies using helium as a carrier gas. FAB-MS analysis was performed with an Ultrahigh Performance Mass Spectrometer JMS-HX110A in Institute for Materials Chemistry and Engineering (IMCE). HPLC analysis was performed on a JASCO LC-2000 Plus Series equipped with a variable wavelength detector using chiral stationary columns (Chiracel AD-H, 0.46 cm x 25 cm) from Daicel. Optical rotations were measured on a JASCO DIP-1000 digital polarimeter. Chromatography was performed on silica-gel (Kanto Chemicals, Silica gel 60N, spherical, neutral; particle size 40–100 µm). Abbreviations: Bn = benzyl, conversion = conv, enantiomeric excess = ee, eq = equiv, er = enantiomer ratio,  $DMAP = N_N$ -dimethylaminopyridine, MTPACl = Methoxy-1-(trifluoromethyl)phenylacetyl chloride, piv = pivaloyl, product = pro, RT = room temperature, substrate = sub.

#### 2. Material.

CDCl<sub>3</sub>, CD<sub>3</sub>OD and C<sub>6</sub>D<sub>6</sub> were used as solvents for NMR analyses. Chloroform was purified prior to use following the guidelines of Perrin and Armarego<sup>1</sup>. Pyridine (anhydrous), CH<sub>2</sub>Cl<sub>2</sub> (anhydrous) and THF (anhydrous, stabilizer free) were used as anhydrous solvents. *N*-9-anthracenylmethyl-cinchonidinium chloride (**1a**, Sigma-Aldrich), *N*-benzylcinchonidinium chloride (**1d**, TOKYO CHEMICAL INDUSTRY CO., LTD.), *N*-benzylcinchoninium chloride (**1e**, TOKYO CHEMICAL INDUSTRY CO., LTD.) and cinchonidine (Wako Pure Chemical Industries, Ltd.) were used as received. All other chemical reagents were used in commercial grade. Catalyst **1b**<sup>2</sup>, **1c**<sup>3</sup>, 2-isopropyl-cycloheptanone<sup>4</sup> (**3h**), 2-allyl-4,4-dimethyl-cyclohexanone<sup>5</sup> and 2-cycloheptylidene-1,1-dimethylhydrazine<sup>6</sup> were prepared according to the reported procedures.

#### **3. Preparation and Characterization of enolesters**

#### 3-1. Synthesis of 2-substituted ketones

#### **4,4-dimethyl-2-propylcyclohexanone**<sup>7</sup> (**3d**)



2-allyl-4,4-dimethylcyclohexanone (570 mg, 3.43 mmol, 1 equiv) in <sup>t</sup>butanol (3.43 mL) was placed in a glass tube with a magnetic stirring bar.  $RuCl_2(PPh_3)_3$  (16.4 mg, 0.017 mmol, 0.5 mol%) was added and the tube was placed in an autoclave. Hydrogen was introduced into the autoclave at a pressure of 2 MPa after hydrogen replacement, then stirred for 11 h at 30 °C. The resultant solution was purified by silica-gel column

chlomatography (Hexane :  $Et_2O = 10:1$ ) to give **3d** (566 mg, 3.36 mmol, 98%) as yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.44$  (td, J = 6.4, 14.2 Hz, 1 H), 2.36 (dt, J = 6.0, 6.0 Hz, 1 H), 2.22 (dt, J = 3.2, 14.2 Hz, 1 H), 1.80-1.55 (m, 4 H), 1.34-1.22 (m, 3 H), 1.19 (s, 3 H), 1.06 (tq, 6.9, 6.9 Hz, 1 H), 0.99 (s, 3 H), 0.87 (t, J = 7.4 Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 214.1$ , 46.8, 45.8, 40.2, 38.6, 31.6, 31.2, 30.9, 24.7, 20.3, 14.3. Anal. Calcd (%) for C<sub>11</sub>H<sub>20</sub>O: C, 78.51; H, 11.98. Found: C, 78.24; H, 11.98.

#### 2-propylcycloheptanone<sup>8</sup>(3f)



BuLi (1.67 M in hexane, 5.25 mmol, 3.14 mL, 1.05 equiv) was added dropwise to a stirred solution of 2-cycloheptylidene-1,1-dimethylhydrazine (771 mg, 5.0 mmol, 1 equiv) in THF (anhydrous, 20 mL) at -5 °C under a nitrogen atmosphere and stirred for 1 h. After that, propyl iodide (893 mg, 512 µL, 5.25 mmol, 1.05 equiv) was added dropwise to the solution, and then the resulting solution was allowed to warm to RT

and stirred for 4 h. Distilled water was added to the resultant solution and cooled to 0 °C. Then the solution was acidified with 1N HCl aq to reach pH 1-2 (The mixture was homogenized by adding THF and methanol). After stirring at 45 °C for another 2 h, the solution was extracted with diethylether. The organic extracts were combined, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The resultant crude product was purified by silica-gel chromatography (hexane:Et<sub>2</sub>O = 20:1) to give **3f** (560 mg, 3.63 mmol, 73%) as pale yellow oil <sup>13</sup>C NMR was in agreement with the literature<sup>8</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.54-2.34 (m, 3 H), 1.89-1.76 (m, 4 H), 1.68-1.48 (m, 2H), 1.40-1.17 (m, 6 H), 0.86 (t, *J* = 7.4 Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  =216.8, 52.3, 42.7, 34.6, 31.3, 29.7, 28.5, 24.8, 20.5, 14.2. Anal. Calcd (%) for C<sub>10</sub>H<sub>18</sub>O: C, 77.87; H, 11.76. Found: C, 78.01; H, 11.83.

### 2-butylcycloheptanone<sup>9,10</sup> (3g)



BuLi (1.67 M in hexane, 5.25 mmol, 3.14 mL, 1.05 equiv) was added dropwise to a stirred solution of 2-cycloheptylidene-1,1-dimethylhydrazine (771 mg, 5.0 mmol, 1 equiv) in THF (anhydrous, 20 mL) at -5 °C under a nitrogen atmosphere and stirred for 1 h. After that, butyl bromide (719 mg, 564  $\mu$ L, 5.25 mmol, 1.05 equiv) was added dropwise to the solution, and then the resulting solution was allowed to warm

to RT and stirred for 4 h. Distilled water was added to the resultant solution and cooled to 0 °C. Then the solution was acidified with 1N HCl aq to reach pH 1-2 (The mixture was homogenized by adding THF and methanol). After stirring at 45 °C for another 2 h, the solution was extracted with diethylether. The organic extracts were combined, dried over Na<sub>2</sub>SO<sub>4</sub> and concentrated. The resultant crude product was purified by silica-gel chromatography (hexane:Et<sub>2</sub>O = 20:1) to give **3g** (643 mg, 3.82 mmol, 77%) as pale yellow oil. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.53-2.32 (m, 3 H), 1.90-1.77 (m, 4 H), 1.68-1.50 (m, 2H), 1.38-1.14 (m, 8 H), 0.86 (t, *J* = 6.9 Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  =216.9, 52.5, 42.7, 32.2, 31.3, 29.7, 29.5, 28.5, 24.8, 22.9, 14.1. Anal. Calcd (%) for C<sub>11</sub>H<sub>20</sub>O: C, 78.51; H, 11.98. Found: C, 78.44; H, 11.96.

#### 3-2. Synthesis of enolesters

**Procedures**: A typical experimental procedure for the preparation of enolesters is described below. Aqueous HClO<sub>4</sub> solution (60%, 106.8  $\mu$ L, 5 mol%) was added to a mixture of 2-propylcyclohexanone **3c** (2.80 g, 20 mmol, 1 equiv) and chloroacetic acid anhydride (6.84 g, 40 mmol, 2 equiv) in CCl<sub>4</sub> (12 mL) and CH<sub>2</sub>Cl<sub>2</sub> (12 mL) under air. The reaction mixture was stirred for 14 h at 30 °C. Then, Et<sub>2</sub>O (50 mL) and saturated NaHCO<sub>3</sub> aq (50 mL) were added to the reaction mixture and stirred for a few minutes. After that, the organic layer was separated and the aqueous layer was extracted with Et<sub>2</sub>O (50 mL x 3). Organic layers were combined and dried with Na<sub>2</sub>SO<sub>4</sub> followed by evaporation. The resultant crude product was purified by silica-gel column chromatography (Et<sub>2</sub>O/Hexane) to give **2c** (3.73 g, 17.2 mmol, 86%) as pale yellow oil.

## 2-ethylcyclohex-1-en-1-yl 2-chloroacetate (2b)

85% yield, yellow oil (4 equiv of 2-chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.12 (s, 2 H), 2.15-2.12 (m, 4 H), 1.93 (q, *J* = 7.9 Hz, 2 H), 1.74-1.66 (m, 2 H), 1.65-1.57 (m, 2 H), 0.92 (t, *J* = 7.8 Hz, 3 H). <sup>13</sup>C NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 165.7, 141.4, 126.7, 40.9, 27.3, 26.9, 23.2, 23.1, 22.4, 12.0. Anal. Calcd (%) for C<sub>10</sub>H<sub>15</sub>ClO<sub>2</sub>: C, 59.26; H, 7.46. Found: C, 59.49; H, 7.47.

## 2-propylcyclohex-1-en-1-yl 2-chloroacetate (2c)

86% yield, yellow oil (2 equiv of chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.11 (s, 2 H), 2.18-2.08 (m, 2 H), 2.08-2.00 (m, 2 H), 1.89 (t, 7.4 Hz, 2 H), 1.74-1.66 (m, 2 H), 1.66-1.56 (m, 2 H), 1.36 (sex, 6.0 Hz, 2 H), 0.85 (t, 7.4Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 165.7, 142.2, 125.2, 40.9, 32.1, 27.8, 26.9, 23.1, 22.4, 20.5, 14.1. Anal. Calcd (%) for C<sub>11</sub>H<sub>17</sub>ClO<sub>2</sub>: C, 60.97; H, 7.91. Found: C, 61.25; H, 8.00.

## 4,4-dimethyl-2-propylcyclohex-1-en-1-yl 2-chloroacetate (2d)

94 %yield, pale yellow oil (3 equiv of 2-chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.12 (s, 2 H), 2.18-2.11(m, 2 H), 1.90-1.80 (m, 4 H), 1.49-1.42 (m, 2 H), 1.39-1.28 (m, 2 H), 0.97-0.92 (m, 6 H), 0.88-0.80 (m, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 165.8, 141.2, 123.9, 41.8, 40.9, 35.6, 32.1, 29.4, 27.9 (2 carbons), 24.5, 20.4, 14.0. Anal. Calcd (%) for C<sub>13</sub>H<sub>21</sub>ClO<sub>2</sub>: C, 63.79; H, 8.65. Found: C, 64.06; H, 8.79.

## [1,1'-bi(cyclohexan)]-1-en-2-yl 2-chloroacetate (2e)

82% yield, yellow oil (2 equiv of chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.11 (s, 2 H), 2.34-2.27 (m, 1 H), 2.13-2.10 (m, 2 H), 2.03-2.00 (m, 2 H), 1.73-1.55 (m, 7 H), 1.46-1.40 (m, 2 H), 1.29-1.03 (m, 5 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 165.8, 140.8, 129.7, 41.0, 38.1, 30.3 (2 carbons), 27.0, 26.6 (2 carbons), 26.2, 23.8, 22.9, 22.5. Anal. Calcd (%) for C<sub>14</sub>H<sub>21</sub>ClO<sub>2</sub>: C, 65.49; H, 8.24. Found: C, 65.64; H, 8.29.

## 2-propylcyclohept-1-en-1-yl 2-chloroacetate (2f)

84% yield, pale yellow oil (4 equiv of chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.09 (s, 2 H), 2.30-2.28 (m, 2 H), 2.11-2.08 (m, 2 H), 1.92 (t, J = 7.3 Hz, 2 H), 1.72-1.50 (m, 6 H), 1.34 (tq, 7.4 Hz, 7.4 Hz, 2 H), 0.84 (t, 7.3 Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 165.9, 146.6, 129.9, 40.9, 34.5, 33.0, 31.5, 31.1, 26.4, 25.3, 20.5, 14.0. Anal. Calcd (%) for C<sub>12</sub>H<sub>19</sub>ClO<sub>2</sub>: C, 62.47; H, 8.30. Found: C, 62.51; H, 8.26.

## 2-butylcyclohept-1-en-1-yl 2-chloroacetate (2g)

85% yield, pale yellow oil (4 equiv of chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.10 (s, 2 H), 2.33-2.27 (m, 2 H), 2.13-2.07 (m, 2 H), 1.95-1.88 (m, 2 H), 1.73-1.65 (m, 2 H), 1.65-1.57 (m, 2 H), 1.57-1.49 (m, 2 H). 1.34-1.29 (m, 4 H), 0.90-0.83 (t, 6.9 Hz, 3 H) <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 166.0, 146.3, 130.2, 40.9, 33.0, 32.1, 31.5, 31.1, 29.5, 26.4, 25.3, 22.6, 14.1. Anal. Calcd (%) for C<sub>13</sub>H<sub>21</sub>ClO<sub>2</sub>: C, 63.79; H, 8.65. Found: C, 64.04; H, 8.62.

### 2-isopropylcyclohept-1-en-1-yl 2-chloroacetate (2h)

56% yield, yellow oil (2 equiv of chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 4.12$  (s, 2 H), 2.69 (sep, 6.9 Hz, 1 H), 2.32-2.24 (m, 2H), 2.08-2.01 (m, 2 H), 1.75-1.65 (m, 2H), 1.65-1.55 (m, 2 H), 1.55-1.45 (m, 2 H), 0.88 (d, 6.9 Hz, 6 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 165.9$ , 144.9, 134.8, 40.9, 33.2, 32.0, 28.2, 26.9, 25.2, 25.1, 20.0(2 carbons). Anal. Calcd (%) for C<sub>12</sub>H<sub>19</sub>ClO<sub>2</sub>: C, 62.47; H, 8.30. Found: C, 62.58; H, 8.28.

## 2-cyclohexylcyclohept-1-en-1-yl 2-chloroacetate (2i)

66% yield, yellow oil (2 equiv of chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.11 (s, 2 H), 2.30-2.27 (m, 3 H), 2.09-2.06 (m, 2 H), 1.73-1.58 (m, 7 H), 1.50-1.40 (m, 4 H), 1.29-1.03 (m, 5 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 166.0, 145.3, 134.5, 40.9, 39.4, 33.2, 32.0, 29.9 (2 carbons), 26.8, 26.37(2 carbons), 26.35, 26.1, 25.3. Anal. Calcd (%) for C<sub>15</sub>H<sub>23</sub>ClO<sub>2</sub>: C, 66.53; H, 8.56. Found: C, 66.70; H, 8.59.

## 2-benzylcyclohept-1-en-1-yl 2-chloroacetate (2j)

96% yield, pale yellow oil (5 equiv of chloroacetic anhydride was used.)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 7.30-7.23$  (m, 2 H), 7.22-7.13 (m, 3 H), 4.11(s, 2 H), 3.30 (s, 2 H), 2.39 (m, 2 H), 2.06-2.01 (m, 2 H), 1.72-1.62 (m, 4 H) 1.45-1.36 (m, 2 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 166.1$ , 147.5, 138.9, 129.0 (2 carbons), 128.6, 128.5 (2 carbons), 126.3, 40.9, 38.2, 33.1, 31.3, 30.8, 26.3, 25.2. Anal. Calcd (%) for C<sub>16</sub>H<sub>19</sub>ClO<sub>2</sub>: C, 68.93; H, 6.87. Found: C, 69.06; H, 6.89.

## 2-(4-methylpentyl)cyclohex-1-en-1-yl 2-chloroacetate (2k)

96% yield, pale yellow oil (4 equiv of chloroacetic anhydride was used)



<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.10 (s, 2 H), 2.16-2.09 (m, 2 H), 2.11-2.07 (m, 2 H), 1.90 (t, J = 7.8 Hz, 2 H), 1.75-1.65 (m, 2 H), 1.65-1.56 (m, 2 H), 1.50 (sep, 6.4 Hz, 1 H), 1.36-1.27 (m, 2 H), 1.15-1.07 (m, 2 H), 0.84 (d, 6.4 Hz, 6 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 165.7, 142.0, 125.5, 40.9, 38.9, 30.3, 27.88, 27.85, 26.9, 25.1, 23.1, 22.7 (2 carbons), 22.4. Anal. Calcd (%) for C<sub>14</sub>H<sub>23</sub>ClO<sub>2</sub>: C, 64.98; H, 8.96. Found: C, 61.12; H, 7.89.

#### 4. General procedure for asymmetric hydrolysis of enolesters catalyzed by PTC

**Procedures**: A typical experimental procedure for asymmetric hydrolysis of enolesters is described below. A round-bottomed screw cap tube ( $\phi$ 13x100 mm) equipped with a magnetic stir bar is charged *N*-9-anthracenylmethyl cinchonidinium chloride (6.1 mg, 0.01 mmol, 10 mol%) and CHCl<sub>3</sub> / Mesitylene (267 µL / 133 µL) solution under air, followed by the addition of 2-chloroethanol (6.7 µL, 0.10 mmol, 1 equiv) and 50 % KOH aq (100 µL). Then, the mixture was stirred at – 40 °C for 10 min followed by the addition of enolester **2c** (21.6 mg, 0.10 mmol, 1 equiv). The efficiency of agitation has an effect on the yield and enantioselectivity. The reaction mixture was stirred for 13 h at – 40 °C. Then, the reaction mixture was immediately passed through a thin pad of silica-gel and the resultant crude product was purified by silica-gel column chlomatography to give (*R*)-**3c** (13.9 mg, 99% yield, 92 : 8 er) followed by Chiral GC analysis (Conditions: InertCap CHIRAMIX, length 30 m, i. D. 0.25 mm, df. 0.25 µm; Detector: FID; Temperature; injector 200 °C, detector 240 °C, oven 40-180 °C, program 3 °C/min, t<sub>major</sub> = 34.7 min, t<sub>minor</sub> = 35.6 min). In case of 1 or 2 mmol-scale and 5 mmol-scale reraction, a 20-mL Schlenk flask and a 50-mL recovery flask were used for an alternative reaction container respectively.

#### 5. Preliminary results of asymmetric hydrolysis of the acetyl enolate (2a).

<u>\_\_\_</u>

**Procedures**: To the stirred solution of **1a** (6.1 mg, 0.1 mmol, 10 mol%) and solid KOH (85% purity, 9.9 mg, 0.15 mmol, 1.5 equiv) was added **2a** (18.2 mg, 0.1 mmol, 1 equiv) at RT. After the 1-14 h, the reaction mixture was directly passed through a short silica-gel pad. Resultant solution was analyzed by GC (according to general procedure for asymmetric hydrolysis of enolesters catalyzed by PTC.). Obtained preliminary results are shown below.

CAC catalyst <b>1a</b> 10 mol% KOH 1.5 equiv solvent, rt					
entry	time (h)	solvent	yield (%) <sup>a)</sup>	er	
1	1	toluene	20	70:30	
2	14	toluene	65	64:36	
3	1	CHCl <sub>3</sub> <sup>b)</sup>	10	79:21	

Table S1 Preliminary results of asymmetric hydrolysis

a) GC yield. b) Less than 1% of EtOH was contained in the solvent.

#### 6-1. Effects of alcohols

**Procedures**: *N*-9-anthracenylmethyl cinchonidinium chloride (6.1 mg, 0.01 mmol, 10 mol%) was added to the CHCl<sub>3</sub> (400  $\mu$ L) under air, followed by the addition of an alcohol (0.05 mmol, 0.5 equiv) and 50% KOH aq (200  $\mu$ L). Then, the mixture was stirred for 10 min at – 40 °C followed by the addition of enolester **2c** (21.6 mg, 0.10 mmol, 1 equiv). The reaction mixture was stirred for 1 h at – 40 °C. Then, the reaction mixture was immediately passed through a thin pad of silica-gel. Resultant solution was analyzed by GC (according to general procedure for asymmetric hydrolysis of enolesters catalyzed by PTC.). Obtained results are shown below.

Table S1 Preliminary results of asymmetric hydrolysis



entry	alcohol	yield (%) <sup>a)</sup>	er	entry	alcohol	yield (%) <sup>a)</sup>	er
1	none	30	85:15	6	PhOH	35	83:17
2	MeOH	64	85:15	7	F₃COH	73	89:11
3	EtOH	66	87:13	8	F OH	42	87:13
4	<sup>/</sup> PrOH	44	84:16	9	CI OH	48	90:10
5	<sup>t</sup> BuOH	2	76:24	10	Br	12	75:15

a) GC yield

#### 6-2. Confirmation of mass balance of reaction products

Asymmetric hydrolysis of 3c was performed according to the general procedures. After 19.5 h, 1 N HCl aq (2 mL) was added to the reaction mixture in order to neutralize KOH aq in the solution. After that, internal standard (diglyme) was added to the solution. The resultant mixture was homogenized by addition of MeOH followed by GC analysis.



6-3. Asymmetric hydrolysis of enolesters with the in situ generated stoichiometric Q<sup>+</sup>OH<sup>-</sup> reagent.

**Reaction in homogenious system (CHCl<sub>3</sub>)**: *N*-9-anthracenylmethyl cinchonidinium chloride (61 mg, 0.10 mmol, 1 equiv) in MeOH solution in 1-neck recovery flask (50 mL) was passed through a column filled with an anion exchange resin (Amberlyst A-26 OH form, 233 mg, 10 meq). MeOH of the eluate was distilled away under reduced pressure at 0 °C. The resultant dried residue was dissolved with CHCl<sub>3</sub> (0.5 mL) and repeatedly dried in vacuo for 15 min at 0 °C. Then, chloroform (1 mL) was added to the dried residue. The mixture was cooled down to -40 °C and stirred for 5 min. An enolester **2c** (21.7 mg,

0.10 mmol, 1 equiv) was added to the solution and stirred for 4 h at the same temperature. After that, the reaction mixture was immediately passed through thin pad of silica-gel and an internal standard (tridecane) was added to the eluate. Then, the resultant mixture was analyzed by GC ((R)-3c, 99%, 89:11 er).

**Reaction in biphasic system (CHCl<sub>3</sub>/H<sub>2</sub>O)**: *N*-9-anthracenylmethyl cinchonidinium chloride (61 mg, 0.10 mmol, 1 equiv) in MeOH solution was passed through a column filled with an anion exchange resin (Amberlyst A-26 OH form, 233 mg, 10 meq). MeOH of the eluate was distilled away under reduced pressure at 0 °C. The resultant dried residue in 1-neck recovery flask (50 mL) was dissolved with CHCl<sub>3</sub> (0.5 mL) and the solution was transferred to a screw vial using MeOH as a solvent. Solvents were removed under reduced pressure at 0 °C. The resultant dried residue was dissolved with CHCl<sub>3</sub> (0.5 mL) and the solution was transferred to a screw vial using MeOH as a solvent. Solvents were removed under reduced pressure at 0 °C. The resultant dried residue was dissolved with CHCl<sub>3</sub> (0.5 mL) and the solution was transferred to a screw vial using MeOH as a solvent. Solvents were removed under reduced pressure at 0 °C. The resultant dried residue was dissolved with CHCl<sub>3</sub> (0.5 mL) and removed repeatedly. Then the dried residue was further dried in vacuo for 15 min at 0 °C. Then, CHCl<sub>3</sub> (400  $\mu$ L) and distilled water (100  $\mu$ L) was added to the residue. The mixture was stirred for 2 min at 25 °C. An enolester **2c** (21.7 mg, 0.10 mmol, 1 equiv) was added to the solution and stirred for 4 h at the same temperature. After that, the reaction mixture was immediately passed through thin pad of silica-gel and an internal standard (tridecane) was added to the eluate. Then, the resultant mixture was analyzed by GC. (*R*)-**3c** (77%, 77:23 er).

# 6-4. Asymmetric hydrolysis of enolesters with the in situ generated stoichiometric $Q^+CF_3CH_2O^-$ reagent.

*N*-9-anthracenylmethyl cinchonidinium chloride (61 mg, 0.10 mmol, 1 equiv) in MeOH solution was passed through a column filled with an anion exchange resin (Amberlyst A-26 OH form, 233 mg, 10 meq). 2,2,2-Trifluoroethanol (125 mg, 1.25 mmol, 90  $\mu$ L, 5 equiv) was added to the elution and stirred for 15 min at RT. The solvent was removed under reduced pressure, then added CHCl<sub>3</sub> and the solution was transferred to a screw vial. Solvents were removed by evaporation and dried in vacuo for 6 h. To the resultant solid was added CHCl<sub>3</sub> (400  $\mu$ L), distilled water (100  $\mu$ L). Then the mixture was stirred for 2 min at 25 °C. An enolester **2c** (21.7 mg, 0.1 mmol, 1 equiv) was added to the solution and stirred for 4 h at the same temperature. After that, the reaction mixture was immediately passed through thin pad of silica-gel and an internal standard (tridecane) was added to the eluate. Then, the resultant mixture was analyzed by GC. (*R*)-**3c** (55%, 69:31 er).

## 6-5. Preparation method for a sample of NOE experiment of *N*-9-anthracenylmethyl cinchonidinium 2,2,2-trifluoroethanoxide.

N-9-anthracenylmethyl cinchonidinium chloride (130 mg, 0.25 mmol, 1 equiv) in MeOH solution was passed through a column filled with an anion exchange resin, Amberlyst A-26 OH form (10 meq). 2, 2, 2-Trifluoroethanol (125 mg, 1.25 mmol, 90  $\mu$ L, 5 equiv) was added to the eluate and stirred for 15 min. The solvent was removed under reduced pressure, then added CDCl<sub>3</sub> (0.5 mL) and removed the solvent again. After that,  $CDCl_3$  (1.0 mL) was added to the residue and 600µL of the solution was transferred to a NMR tube. D<sub>2</sub>O (50  $\mu$ L) was added to the sample. After that, it was subjected to two freeze-pumpthaw cycles followed by keeping in refrigerator for 12 h. Then, NMR experiments were performed.<sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta = 8.79$  (d, J = 8.9 Hz, 1 H, H<sub>19</sub>), 8.64 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H<sub>1</sub>), 8.61 (d, J = 3.4 Hz, 1 H, H\_1), 8.61 (d, J = 3.4 Hz, 1 H, H\_1), 8.61 (d, J = 3.4 8.2 Hz, 1 H, H<sub>22</sub>), 7.70 (d, J = 8.2 Hz, 1 H, H<sub>26</sub>), 7.47 (d, J = 8.3 Hz, 1 H, H<sub>8</sub>), 7.40-7.28 (m, 3 H, H<sub>20</sub> +  $H_{21} + H_{28}$ , 7.28-7.23 (m, 1 H,  $H_{27}$ ), 7.06 (t, J = 6.8 Hz, 1H,  $H_6$ ), 6.99-6.94 (m, 1 H,  $H_7$ ), 6.98 (s, 1H,  $H_{10}$ ), 6.46 (d, J = 13.7 Hz, 1 H,  $H_{16b}$ ), 6.41 (d, J = 13.7 Hz, 1 H,  $H_{16a}$ ), 5.36 (ddd, J = 6.6, 11.0, 17.2 Hz, 1 H, H<sub>33</sub>), 5.01 (d, J = 17.2 Hz, 1 H, H<sub>34a</sub>), (d, J = 11.0 Hz, 1 H, H<sub>34b</sub>), 4.76-4.66 (m, 1 H, H<sub>15b</sub>), 4.76-4.66 (m, 1 H, H<sub>15b</sub>), 4.49-4.43 (m, 1 H, H<sub>11</sub>), 4.05-3.93 (m, 3.1H, H<sub>35</sub> + excess), 3.92-3.83 (m, 1 H, H<sub>32b</sub>), 2.61 (t, J = 12.4 Hz,  $H_{32b}$ ), 2.65-2.57 (m, 1 H,  $H_{32a}$ ), 2.40-2.31 (m, 1 H,  $H_{15a}$ ), 1.66-1.61 (m, 1 H,  $H_{31}$ ), 1.90-1.72 (m, 2 H, H<sub>14b</sub> + H<sub>12b</sub>), 1.66-1.61 (m, 1H, H<sub>13</sub>), 1.11-1.02 (m, 1 H, H<sub>14a</sub>), 0.95-0.87 (m, 1 H,  $H_{12a}$ ).



Green: negative, Red: positive











#### **2d-ROESY Chart**



Green: negative, Red: positive

## 6-6. Kinetic resolution of an aryl ester bearing binaphthyl backbone

## 6-6-1. Synthesis of the catalyst and substrate



#### Synthesis of 1-bromomethyl-2,3,4,5-tetraphenyl benzene (8)

To a stirred solution of tetraphenyl cyclopentadienone (3.85 g, 10 mmol, 1 equiv) in toluene (10 mL) was added propargyl bromide (3.01 mL, 40 mmol, 4 equiv). After stirring for 8 h at 110 °C the reaction mixture was allowed to cool down to RT. Resultant white precipitate was filtrated and washed with hexane to give **8** (4.36 g, 9.1 mmol, 91%) as white solid. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.64 (s, 1 H), 7.20-7.11 (m, 10 H), 6.92-6.88 (m, 3 H), 6.86-6.81 (m, 3 H), 6.81-6.77 (m, 2 H), 6.76-6.72(m, 2 H), 4.40 (s, 2 H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta$  = 142.2, 141.5, 141.4, 140.9, 140.6, 139.8, 139.6, 138.7, 135.1, 131.5, 131.4(2 carbons), 131.2 (2 carbons), 130.3 (2 carbons), 129.9 (2 carbons), 127.7 (2 carbons), 127.6 (2 carbons), 127.0 (2 carbons), 126.8, 126.7 (2 carbons), 126.5, 125.8, 125.6, 32.8. Anal. Calcd (%) for C<sub>31</sub>H<sub>23</sub>Br: C, 78.32; H, 4.88. Found: C, 78.29; H, 4.86.

#### Synthesis of catalyst 1g

A stirred solution of quinine (1 mmol, 324 mg, 1 equiv) in DMF:EtOH:CHCl<sub>3</sub> (9:7.5:1, 200  $\mu$ L) solution was added 1-bromomethyl-2,3,4,5-tetraphenyl benzene **8** (1 mmol, 475 mg, 1 equiv). The mixture was allowed to warm up to 100 °C. After 2 h, the reaction mixture was cooled to RT. The resultant solution was evaporated and purified by Silica-gel column chromatography to give **1g** (344 mg, 0.43 mmol, 43%) as pale pink powder. <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD)  $\delta$  =8.73-8.67 (m, 1 H), 7.97 (d, *J* = 9.2 Hz, 1 H), 8.00-7.95 (m, 2 H), 7.77 (d, *J* = 4.6 Hz, 1H) 7.48 (dd, *J* = 2.3, 9.6 Hz, 1 H), 7.31-7.10 (m, 8 H), 6.91-6.71 (m, 12 H), 6.49 (s, 1 H), 5.63-5.52 (m, 1H), 5.59 (d, *J* = 12.4 Hz, 1 H), 4.96-4.84 (m, 3 H), 4.05 (s, 3 H), 3.95-3.85 (m, 1 H), 3.78 (t, *J* = 11.4 Hz, 1 H), 3.64-3.55 (m, 1 H), 3.55-3.44 (m, 1 H), 3.28 (s, 1 H), 3.04-2.94 (m, 1 H), 2.76 (s, 1 H), 2.16-2.24 (m, 2 H), 1.98 (s, 1 H), 1.95-1.84 (m, 1 H), 1.31-1.20 (m, 1 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 158.7, 146.9, 144.5, 144.0, 143.6, 143.3, 143.1, 141.7, 140.8, 139.6, 139.2, 138.6, 137.5, 135.6, 131.6, 131.4, 131.04, 131.00, 130.94, 130.6, 129.7, 128.1, 128.0, 127.6, 127.1, 126.8, 126.6, 125.9, 125.6, 124.4, 121.4, 120.2, 115.8, 101.4, 68.4, 64.9, 61.8, 61.1, 55.4, 53.5, 51.6, 37.6, 26.3, 24.5, 21.2. HRMS (FAB+) m/z calculated for C<sub>51</sub>H<sub>47</sub>N<sub>2</sub>O<sub>2</sub> 719.3632 found 719.3641 [M-Br]

#### Synthesis of 2'-acetoxy-[1,1'-binaphthalen]-2-yl pivalate (7)



To a mixture of 2'-hydroxy-[1,1'-binaphthalen]-2-yl pivalate (1.14 g, 4.0 mmol) and acetyl chloride (310  $\mu$ L, 4.4 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) was added Et<sub>3</sub>N (610  $\mu$ L, 4.4 mmol) at 0 °C and the mixture was stirred at RT for 20 h. The reaction was quenched by adding HCl (1N, 30 mL) and extracted with AcOEt. The resulting extracts were washed with brine, and dried, and concentrated under reduced pressure. The crude product was purified by silica-gel column chromatography to give **6** (1.63 g, 3.96

mmol, 99%) as white solid. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>)  $\delta = 7.99-7.95$  (m, 1 H), 7.93-7.89 (m, 1 H), 7.46-7.37 (m, 4 H), 7.31-7.22 (m, 6 H), 1.76 (s, 3 H), 0.76 (s, 9 H). <sup>13</sup>C-NMR (100 MHz, CDCl<sub>3</sub>)  $\delta = 176.6, 169.2, 147.0, 146.9, 133.5, 133.4, 131.6, 131.5, 129.5, 129.4, 128.1, 128.0, 126.8 (2 carbons), 126.2, 126.1, 125.8, 125.7, 123.7, 123.6, 122.0, 121.9, 38.8, 26.5 (3 carbons), 20.6. elemental analysis calcd (%) for C<sub>27</sub>H<sub>24</sub>O<sub>4</sub>: C 78.62, H 5.86; found: C 78.60, H 5.86.$ 

#### 6-6-2. Asymmetric hydrolysis of the acetate ester

A mixture of the catalyst **1g** (0.01 mmol, 10 mol%) and 1N K<sub>2</sub>CO<sub>3</sub> aq (200 µL, 0.2 mmol, 2 equiv) in toluene (200 µL) was stirred at 20 °C. After 20 min, the *rac*-**7** (0.1 mmol, 1 equiv) was added and the mixture was stirred at the same temperature. Silica-gel column chromatography (Hexane: CHCl<sub>3</sub>: Et<sub>2</sub>O = 20:6.5:1) was carried out to give (*S*)-**8** (15% yield, 78:22 er) and (*R*)-**7** (80% yield, 43.5:56.5 er); HPLC analysis: CHIRALPAK AD-H (+), hexane/2-propanol = 20/1, flow rate = 1.0 mL/min, 25 °C, **7**:  $t_{major} = 7.9$  min,  $t_{minor} = 17.4$  min, **8**:  $t_{major} = 10.6$  min,  $t_{minor} = 16.5$  min)(*S*)-**7**:  $[\alpha]_D^{25.5} = +35.1$ , (*c* 0.5, CHCl<sub>3</sub>), (*R*)-**8**:  $[\alpha]_D^{24.5} = -3.63$ , (*c* 1.0, CHCl<sub>3</sub>).  $k_{rel} = 4.1$  (The  $k_{rel}$  value was calculated from ee<sub>sub</sub> and ee<sub>pro</sub> with equations as follows.: conv = ee<sub>sub</sub>/(ee<sub>sub</sub>+ee<sub>pro</sub>),  $k_{rel} = \ln[(1 - \text{conv} (1 + ee_{pro})]/\ln[(1 - \text{conv} (1 - ee_{pro})])$ 

#### **Recovered substrate** (S)-7



<sup>1</sup>H NMR (600 MHz, CDCl3)  $\delta$  = 7.99-7.95 (m, 1 H), 7.93-7.89 (m, 1 H), 7.46-7.37 (m, 4 H), 7.31-7.22 (m, 6 H), 1.76 (s, 3 H), 0.76 (s, 9 H). 13C-NMR (100 MHz, CDCl3)  $\delta$  = 176.6, 169.2, 147.0, 146.9, 133.5, 133.4, 131.6, 131.5, 129.5, 129.4, 128.1, 128.0, 126.8 (2 carbons), 126.2, 126.1, 125.8, 125.7, 123.7, 123.6, 122.0, 121.9, 38.8, 26.5 (3 carbons), 20.6. elemental analysis calcd (%) for C<sub>27</sub>H<sub>24</sub>O<sub>4</sub>: C 78.62, H 5.86; found: C 78.60, H 5.86.

#### Product (*R*)-8



<sup>1</sup>H and <sup>13</sup>C NMR were in agreement with the literature<sup>11</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>)  $\delta = 8.06$  (d, J = 6.0 Hz, 1 H), 7.97 (d, J = 5.5 Hz, 1 H), 7.87 (d, J = 6.0 Hz, 1 H), 7.81 (d, J = 5.5 Hz, 1H), 7.50(t, J = 5.5 Hz, 1H), 7.39-7.28 (m, 5H), 7.26-7.21 (m, 1 H), 7.05 (d, J = 5.5 Hz, 1 H), 5.14 (s, 1 H), 0.78 (s, 9 H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>)  $\delta = 177.9$ , 151.8, 148.4, 133.7, 133.6, 132.3, 130.8, 130.4, 129.1, 128.4, 128.0, 127.5, 126.7, 126.3, 125.7, 124.6, 123.6, 123.1, 121.9, 118.3, 114.3, 38.8, 140.4 Hz = 0.04.06 Hz = 0.

26.5(3 carbons). elemental analysis calcd (%) for C<sub>25</sub>H<sub>22</sub>O<sub>3</sub>: C, 81.06; H, 5.99; found: C, 81.11, H, 5.99.



HPLC chart of recovered substrate (S)-7 and product (R)-8

# 7. Formal synthesis of the biologically-active natural product 7-1. Synthesis of catalyst 1f



#### Synthesis of amine 9

To the stirred solution of (-)-cinchonidine (883 mg, 3 mmol, 1 equiv) in dry Et<sub>2</sub>O (15 mL) at -25 °C was added <sup>7</sup>BuLi (1.57M in pentane, 5.73 mL, 9 mmol, 3 equiv) in one portion and stirred for 10 min. Then, the reaction mixture was allowed to warm up to RT and stirred for another 1 h. Reaction was monitored by TLC analysis (toluene:MeOH:Et<sub>3</sub>N = 10:1:1, <sup>t</sup>Bu adducts turn blue under the UV irradiation). AcOH was added to quench the residual basic reagents in cool bath, then EtOAc (30 mL) and H<sub>2</sub>O (30 mL) was added followed by the addition of I<sub>2</sub> until strong brown color persists. After that,  $Na_2S_2O_3$  aq was added to quench residual I<sub>2</sub>. The reaction mixture was extracted with EtOAc. The organic layer was dried with Na<sub>2</sub>SO<sub>4</sub> and evaporated. The crude mixture was purified by silica-gel column chromatography (toluene:MeOH:Et<sub>3</sub>N = 20:1:1) to give the product 9 (704 mg, 1.92 mmol, 64%) as white solid. Analytical sample was prepared by the PTLC purification ( $CH_2Cl_2$ :MeOH = 4:1) <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 8.07$  (d, J = 8.2 Hz, 1 H), 7.90 (d, J = 8.7 Hz, 1 H), 7.72, (s, 1 H), 7.63 (t, J = 7.8 Hz, 1H), 7.39 (t, J = 7.4 Hz, 1H), 5.71 (ddd, J = 7.8, 10.1, 17.4 Hz, 1 H), 5.66 (s, 1 H), 4.93 (d, J = 17.4 Hz, 1 H), 4.88 (d, J = 10.1 Hz, 1 H), 3.53-3.43 (m, 1 H), 3.12-3.00 (m, 2 H), 2.70-2.59 (m, 2 H), 2.29-2.20 (m, 1 H), 1.80-1.65 (m, 4 H), 1.50-1.40 (m, 11 H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>): δ = 169.0, 148.3, 147.7, 142.0, 130.4, 128.7, 125.9, 123.9, 122.6, 115.2, 114.4, 72.4, 60.3, 57.2, 43.4, 40.1, 38.3, 30.2 (3 carbons), 28.1, 27.7, 21.4. FABMS m/z calculated for C<sub>23</sub>H<sub>30</sub>N<sub>2</sub>O 351.24 found 351.29 [M+H] Anal. Calcd (%) for C<sub>23</sub>H<sub>30</sub>N<sub>2</sub>O: C, 78.82; H, 8.63; N, 7.99 Found: C, 78.38; H, 8.63, N; 7.89.

#### Synthesis of 1f

To the stirred solution of amine 9 (73 mg, 0.2 mmol, 1 equiv) in CHCl<sub>3</sub>-THF (1:1) was added 9bromomethylanthracene (57 mg, 0.21 mmol, 1.05 equiv) followed by concentration with nitrogen gas stream down to a volume 0.2 mL. The mixture was stirred at 80 °C for 10 min. Then, the solution was allowed to cool down to RT. After that, resultant precipitate was dissolved with CHCl<sub>3</sub>. To the solution was added Et<sub>2</sub>O dropwise to solidify the product. The resulting solid was filtrated and washed with Et<sub>2</sub>O to give the product (118 mg, 0.19 mmol, 95%) as pale yellow powder. Analytical sample was prepared by the PTLC purification (EtOAc:MeOH = 4:1). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  = 8.81 (d, J = 8.9 Hz, 1 H), 8.68 (s, 1 H), 8.60-8.54 (m, 1 H), 8.54-8.48 (m, 1 H), 8.16-8.09 (m, 4 H), 7.82-7.70 (m, 4 H), 7.60-7.52 (m, 2 H), 7.01 (s, 1 H), 6.36 (d, J = 13.7 Hz, 1 H), 5.86 (d, J = 13.7 Hz, 1 H), 5.66 (ddd, J = 7.6, 9.6, 9.6, 10.517.2 Hz, 1 H), 5.00 (d, J = 17.2 Hz, 1 H), 4.92 (d, J = 9.6 Hz, 1 H), 4.66-4.59 (m, 1 H), 4.48-4.40 (m, 1 H), 3.90-3.84 (m, 1 H), 3.13 (t, J = 11.7 Hz, 1 H), 2.76-2.66 (m, 1 H), 2.36 (br s, 1 H), 2.24-2.14 (m, 1 H), 2.12-2.00 (m, 1 H), 1.87 (s, 1 H), 1.66-1.20 (s, 9 H), 1.45-1.37 (m, 1 H), 1.35-1.20 (m, 1 H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>):  $\delta = 169.0, 147.5, 145.1, 137.5, 133.4, 133.3, 132.3, 131.60, 131.55, 129.8, 133.4, 133.3, 132.3, 131.60, 131.55, 129.8, 133.4, 133.4, 133.4, 133.4, 133.4, 134$ 129.7, 129.6, 129.2, 127.93, 127.88, 126.9, 125.2, 124.5, 123.9, 123.1, 122.7, 118.0, 117.0, 116.3, 78.2, 68.6, 66.2, 62.3, 55.4, 51.9, 38.3, 38.0, 29.2 (3 carbons), 26.0, 24.9, 21.8. FABMS m/z calculated for C<sub>38</sub>H<sub>41</sub>N<sub>2</sub>O<sup>+</sup> 541.32 found 541.38 [M].

#### Synthesis of (R)-3k from cyclohexanone (4)



To a stirred solution of cyclohexanone (4.91 g, 50 mmol, 1 equiv) and N, N-dimethyl hydrazine (3.16 g, 52.5 mmol, 1.05 equiv) in toluene (40 mL) was added trifluoroacetic acid (80 mg, 0.7 mmol, 1.4 mol%). The mixture was refluxed for 5 h. After that, the resulting solution was cooled to RT. Then, distilled water was added to the solution and extracted with  $Et_2O$ . The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub>. The solution was evaporated and purified by distillation under reduced pressure to give colorless oil (yield of 7: 6.13 g, 43.7 mmol, 87%) laced with toluene (molar ratio = 10 : 1). This was used without further purification. BuLi (1.67 M in hexane, 10.5 mmol, 6.30 mL, 1.0 equiv) was added dropwise to a stirred solution of 7 (1.47 g, 10.5 mmol, 1 equiv) in THF (anhydrous, 20 mL) at -5 °C under a nitrogen atmosphere and stirred for 1 h. After that, 1-bromo-4-methylpentane (1.733 g, 10.5 mmol, 1 equiv) was added dropwise to the solution, and then the resulting solution was allowed to warm to RT and stirred for 5 h. The resultant solution was added water and cooled to 0 °C. Then the solution was acidified with concentrated HCl aq to reach pH = 1-2. After stirring at 45 °C for another 3 h, the solution was extracted with Et<sub>2</sub>O. The organic extracts were combined, dried over Na<sub>2</sub>SO<sub>4</sub>, and concentrated to give yellow oil. Silica-gel column chromatography of the crude residue by eluting with diethylether/hexane (1:30-1:20) afforded **3k** (1.76 g, 9.66 mmol, 92%) as a colorless oil and <sup>1</sup>H NMR was in agreement with the literature<sup>12</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.40-2.32$  (m, 1 H), 2.32-2.18 (m, 2H), 2.13-2.04 (m, 1 H), 2.04-1.92 (m, 1 H), 1.88-1.79 (m, 1 H), 1.79-1.57 (m, 3 H), 1.51 (sep, J = 6.4 Hz, 1 H), 1.42-1.31 (m, 1H), 1.31-1.08 (m, 5 H), 0.84 (d, J = 6.9 Hz, 6 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 213.8$ , 50.9, 42.1, 39.1, 33.9, 29.7, 28.1, 27.9, 25.0, 24.9, 22.71, 22.66. Anal. Calcd (%) for C12H22O: C, 79.06; H, 12.16. Found: C, 79.03; H, 12.12.



**2k** was synthesized according to the general method (pale yellow oil, 96% yield). Asymmetric hydrolysis of **2k** was performed according to the general method of Asymmetric hydrolysis of enolesters (cat. **1f** was used in place of cat. **1a**). (R)-(+)-2-(4-Methylpentyl)cycloheptanone was obtained in 96% yield and 92: 8 er as a colorless oil.

#### 8. Derivatization of ketones to the corresponding alcohol and Mosher's esters



#### General procedure: reduction of ketones

A typical experimental procedure for the reduction of ketones is described below. To the stirred solution of a ketone (0.01 mmol, 1 equiv) in absolute EtOH (0.5 mL) was added NaBH<sub>4</sub> (9.5 mg, 0.25 mmol, 2.5 equiv) followed by stirring at RT for 24-48 h. The reaction mixture was monitored by TLC. After the substrate was completely converted, the reaction mixture was quenched with water and 1N HCl aq. The resultant solution was extracted with  $Et_2O$  (x 3). After that, the organic layer was combined and dried over Na<sub>2</sub>SO<sub>4</sub>. The crude product was concentrated and purified by silica-gel chromatography to give the corresponding alcohol.

### 2-propyl-cycloheptanol (10f)<sup>13</sup>



0.0758 mmol (11.7 mg) of 2-propylcyclohepanone **3f** (Table 2, entry 5) was used. The product (diastereomeric mixture) was obtained as colorless oil (11.5 mg, 0.0736 mmol, 97%). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta = 3.94$ -3.89 (m, 0.66 H), 3.49-3.44 (m, 0.34 H), 1.80-1.15 (m, 16 H), 0.93-0.86 (m, 3H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>): major isomer,  $\delta = 73.4$ , 44.2, 35.6, 35.2, 28.3, 27.2, 26.8, 22.0, 21.0, 14.5. minor 47.2, 26.8, 26.5, 20.1, 28.7, 26.0, 22.3, 20.2, 14.6

isomer,  $\delta = 73.4, 47.2, 36.8, 36.5, 29.1, 28.7, 26.9, 22.3, 20.2, 14.6$ .

## 2-butyl-cycloheptanol (10g)<sup>9,14</sup>



0.097 mmol (16.3 mg) of 2-butylcycloheptanone **3g** (Table 2, entry 6) was used. The product (diastereomixture) was obtained as colorless oil (0.0916 mmol, 15.6 mg, 94%). <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  = 3.94-3.89 (m, 0.7 H), 3.71-3.20 (m, 0.3 H), 1.80-1.10 (m, 18 H), 0.95-0.80 (m, 3 H). <sup>13</sup>C NMR (150 MHz, CDCl<sub>3</sub>): major isomer,  $\delta$  = 73.4, 44.5, 35.6, 32.6, 30.2, 28.3, 27.2, 26.85, 23.1, 22.0, 14.2.

minor isomer,  $\delta = 73.4, 47.4, 36.5, 34.2, 29.4, 29.1, 28.8, 26.91, 23.2, 22.3, 14.2$ .

## 2-isopropyl-cycloheptanol(10h)<sup>15</sup>



0.097 mmol (15 mg) of 2-isopropylcycloheptanone **3h** (Table 2, entry 7) was used. The product was obtained as colorless oil in 93% yield (major isomer: 10.8 mg, 0.0691 mmol, 71%, minor isomer: 3.3 mg, 0.0211 mmol, 22%). <sup>1</sup>H NMR of major isomer (600 MHz, CDCl<sub>3</sub>):  $\delta$  = 4.13-4.09 (m, 1 H), 1.76-1.67 (m, 3 H), 1.67-1.49 (m, 5 H), 1.48-1.35 (m, 3 H), 1.19-1.13 (m, 2 H), 0.95 (t, *J* = 5.5 Hz, 6 H). <sup>13</sup>C NMR of major

isomer (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 71.5, 50.2, 36.9, 31.4, 28.2, 27.9, 24.1, 22.2, 21.1. Anal. Calcd (%) for C<sub>10</sub>H<sub>20</sub>O: C, 76.86; H, 12.90. Found: C, 76.44; H, 12.64.

#### Mosher's esterification

### (2R)-2-propylcycloheptyl 3,3,3-trifluoro-2-methoxy-2-phenylpropanoate (11f)



To the stirred solution of an alcohol (0.0644 mmol, 1 equiv) in dry  $CH_2Cl_2$  (0.5 mL) was added (*R*)-MTPACl (32.5 mg, 0.129 mmol, 2.0 equiv), dry pyridine (20.4 mg, 0.258 mmol, 4 equiv) and *N*,*N*-dimethyl aminopyridine (1 mg, 4 µmol, 6 mol%) followed by stirring at RT for 48 h. The reaction mixture was monitored by TLC. After the substrate was completely converted, the reaction mixture was diluted with Et<sub>2</sub>O. The crude was washed with saturated CuSO<sub>4</sub>

aq and extracted with Et<sub>2</sub>O (x3). The resultant solution was was concentrated and purified by silica-gel chromatography (Hexane/Et<sub>2</sub>O = 50:1-40:1) to give the corresponding ester (93%) as colorless oil. <sup>1</sup>H NMR (600 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.56-7.49 (m, 2 H), 7.41-7.35 (m, 3 H), 5.32-5.29 (m,0.335 H), 5.29-5.24 (m, 0.335 H), 4.96-4.88 (m, 0.33 H), 3.57-3.50 (m, 3 H), 1.95-1.00 (m, 15 H), 0.90-0.71 (m, 3 H).

#### (2R)-2-butylcycloheptyl 3,3,3-trifluoro-2-methoxy-2-phenylpropanoate(11g)



To the stirred solution of an alcohol (0.059 mmol, 1 equiv) in dry  $CH_2Cl_2$  (0.5 mL) was added (*R*)-MTPACl (29.8 mg, 0.118 mmol, 2.0 equiv), dry pyridine (18.7 mg, 0.16 mmol, 4 equiv) and *N*,*N*-dimethyl aminopyridine (1mg, 4 µmol, 7 mol%) followed by stirring at RT for 48 h. The reaction mixture was monitored by TLC. After the substrate was completely converted, the reaction mixture was diluted with Et<sub>2</sub>O. The crude was washed with saturated CuSO<sub>4</sub>

aq and extracted with Et<sub>2</sub>O (x3). The resultant solution was was concentrated and purified by silica-gel chromatography (Hexane/Et<sub>2</sub>O = 50:1-40:1) to give the corresponding ester (99%) as inhomogeneous colorless oil. <sup>1</sup>H NMR (600 MHz, C<sub>6</sub>D<sub>6</sub>):  $\delta$  = 7.71-7.63 (m, 2 H), 7.08-7.01 (m, 2 H), 7.01-6.95 (m, 1 H), 5.28-5.22 (m, 0.72 H), 4.96-4.93 (m, 0.14 H), 4.92-4.88 (m, 0.14 H), 3.43-3.39 (m, 3 H), 1.83-0.74 (m, 20 H).

#### 9. Analysis of hydrolyzed products



The product ((*S*)-(+)-2-ethyl-cyclohexanone) was obtained as a colorless oil and 87: 13 er . <sup>1</sup>H and <sup>13</sup>C NMR were in agreement with the literature<sup>16</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.40$ -1.90 (m, 5 H), 1.90-1.54 (m, 4 H), 1.44-1.29 (m, 1 H), 1.29-1.15 (m, 1 H), 0.86 (t, J = 7.3 Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 213.6$ , 52.4, 42.1, 33.5,

28.1, 24.9, 22.5, 11.8. Anal. Calcd (%) for C<sub>8</sub>H<sub>14</sub>O: C, 76.14; H, 11.18. Found: C, 76.30; H, 11.21. Enantiomeric ratio (er) was determined by GC with a CHIRASIL-DEX CB column (conditions, starting temperature: 60 °C [hold 10 min.], rate of temperature increase: 2 °C/min up to 120 °C), t<sub>r</sub> (major) = 27.6 min., t<sub>r</sub> (minor) = 28.3 min.[ $\alpha$ ]<sub>D</sub><sup>25.6</sup> = + 36.1, (*c* 0.5, CHCl<sub>3</sub>) The absolute configuration was established by comparison of the optical rotation to the literature value for (*R*)-(–)-2-ethylcyclohexanone: [ $\alpha$ ]<sub>D</sub><sup>25</sup> = – 23.6 (*c* 4.31, MeOH)<sup>17,18</sup>.



The product ((*S*)-(+)-2-propylcyclohexanone) was obtained as a pale yellow oil and 92: 8 er. <sup>1</sup>H and <sup>13</sup>C NMR were in agreement with the literature<sup>19</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.39-2.2.31$  (m, 1 H), 2.31-2.20 (m, 2H), 2.12-1.92 (m, 2 H), 1.86-1.56 (m, 4 H), 1.42-1.08 (m, 4 H), 0.87 (t, *J* = 6.9 Hz, 3 H). <sup>13</sup>C NMR (100 CMR) (100

MHz, CDCl<sub>3</sub>):  $\delta = 213.7$ , 50.6, 42.0, 33.9, 31.7, 28.1, 24.9, 20.4, 14.3. Anal. Calcd (%) for C<sub>9</sub>H<sub>16</sub>O: C, 77.09; H, 11.50. Found: C, 77.01; H, 11.50. Enantiomeric ratio (er) was determined by GC with a InertCap CHIRAMIX Column (conditions, starting temperature: 40 °C [hold 0 min.], rate of temperature increase: 3 °C/min up to 120 °C [hold 15 min.]), t<sub>r</sub> (major) = 37.5 min., t<sub>r</sub> (minor) = 38.7 min.[ $\alpha$ ]<sub>D</sub><sup>22.5</sup> = + 33.5, (*c* 1.0, CHCl<sub>3</sub>) The absolute configuration was established by comparison of the

optical rotation to the literature value for (*R*)-(–)-2-propylcyclohexanone:  $[\alpha]_D^{25} = -25.7$  (*c* 0.82, MeOH)<sup>20</sup>.



The product<sup>7</sup> ((+)-4,4-dimethyl-2-propylcyclohexanone) was obtained as a pale yellow oil and 92: 8 er. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.44$  (td, J = 6.4, 14.2 Hz, 1 H), 2.36 (dt, J = 6.0, 6.0 Hz, 1 H), 2.22 (dt, J = 3.2, 14.2 Hz, 1 H), 1.80-1.55 (m, 4 H), 1.34-1.22 (m, 3 H), 1.19 (s, 3 H), 1.06 (tq, 6.9, 6.9 Hz, 1 H), 0.99 (s, 3 H), 0.87 (t, J = 7.4 Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 214.1$ , 46.8, 45.8, 40.2, 38.6, 31.6,

31.2, 30.9, 24.7, 20.3, 14.3. Anal. Calcd (%) for  $C_{11}H_{20}O$ : C, 78.51; H, 11.98. Found: C, 78.24; H, 11.98. Enantiomeric ratio (er) was determined by GC with a InertCap CHIRAMIX Column (conditions, starting temperature: 40 °C [hold 0 min.], rate of temperature increase: 3 °C/min up to 120 °C [hold 15 min.]),  $t_r$  (major) = 42.9 min.,  $t_r$  (minor) = 44.1 min.  $[\alpha]_D^{25.9} = + 26.6$ , (*c* 1.0, CHCl<sub>3</sub>)



The product ((*R*)-(+)-2-cyclohexylcyclohexanone) was obtained as a colorless oil and 87: 13 er and <sup>1</sup>H and <sup>13</sup>C NMR were in agreement with the literature<sup>12</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.39-2.29$  (m, 1 H), 2.28-2.18 (m, 1 H), 2.11-2.01 (m, 1 H), 1.98-1.45 (m, 12 H), 1.33-1.18 (m, 2 H), 1.16-0.80 (m, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 213.8$ , 56.6, 41.9, 36.1, 31.6, 29.42, 29.37, 28.0, 26.6, 26.5, 24.0. Anal.

Calcd (%) for C<sub>12</sub>H<sub>20</sub>O: C, 79.94; H, 11.18. Found: C, 79.78; H, 11.27. Enantiomeric ratio (er) was determined by GC with a CHIRASIL-DEX CB column (conditions, starting temperature: 40 °C [hold 1 min.], rate of temperature increase: 2 °C/min up to 160 °C),  $t_r$  (minor) = 55.1 min.,  $t_r$  (major) = 55.3 min.  $[\alpha]_D^{24.8} = +$  46.1, (*c* 1.0, CHCl<sub>3</sub>) The absolute configuration was established by comparison of the

optical rotation to the literature value for (S)-(-)-2-cyclohexylcyclohexanone:  $\left[\alpha\right]_{D}^{24.3} = -38.1$  (c 1.58,  $MeOH)^{21}$ .





The product (2-(+)-propylcycloheptanone) was obtained as a pale yellow oil and 94.5: 5.5 er and <sup>13</sup>C NMR was in agreement with the literature<sup>8</sup>. <sup>1</sup>H NMR (400 MHz,  $CDCl_3$ ):  $\delta = 2.54-2.34$  (m, 3 H), 1.89-1.76 (m, 4 H), 1.68-1.48 (m, 2H), 1.40-1.17 (m, 6 H), 0.86 (t, J = 7.4 Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 216.8, 52.3, 42.7,$ 34.6, 31.3, 29.7, 28.5, 24.8, 20.5, 14.2. Anal. Calcd (%) for C<sub>10</sub>H<sub>18</sub>O: C, 77.87; H, 11.76. Found: C, 78.01; H, 11.83.  $[\alpha]_D^{25.3} = +49.7$  (c 1.0, CHCl<sub>3</sub>). Enantiomeric ratio (er) was

determined by <sup>1</sup>H NMR after the product was reduced and esterified to the corresponding mosher's ester (600 MHz, C<sub>6</sub>D<sub>6</sub> major isomer :  $\delta = 5.33-5.29$  ppm, minor isomer :  $\delta = 5.29-5.25$  ppm).



The product  $^{9}(2-(+)-butylcycloheptanone)$  was obtained as a pale yellow oil and 95: 5 er. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.53-2.32$  (m, 3 H), 1.90-1.77 (m, 4 H), 1.68-1.50 (m, 2H), 1.38-1.14 (m, 8 H), 0.86 (t, J = 6.9 Hz, 3 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ =216.9, 52.5, 42.7, 32.2, 31.3, 29.7, 29.5, 28.5, 24.8, 22.9, 14.1. Anal. Calcd (%) (%) for C<sub>11</sub>H<sub>20</sub>O: C, 78.51; H, 11.98. Found: C, 78.44; H, 11.96.

 $\left[\alpha\right]_{D}^{25.3} = +49.7$  (c 1.0, CHCl<sub>3</sub>) Enantiomeric ratio (er) was determined by <sup>1</sup>H NMR after the product was reduced and esterified to the corresponding mosher's ester (600 MHz,  $C_6D_6$  major isomer :  $\delta =$ 4.97-4.92 ppm, minor isomer :  $\delta = 4.92-4.87$  ppm).



The product ((*R*)-(+)-2-benzylcycloheptanone) was obtained as a colorless oil and 89.5: 10.5 er and <sup>13</sup>C NMR was in agreement with the literature<sup>22</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.29-7.22 (m, 2 H), 7.21-7.11 (m, 3 H), 3.06 (dd, *J* = 6.0, 13.8 Hz, 1 H), 2.86-2.75 (m, 1 H), 2.54 (dd, *J* = 8.7, 13.8 Hz, 1 H), 2.48-2.40 (m, 2H), 1.90-1.70 (m, 4 H), 1.68-1.54 (m, 1 H), 1.38-1.22 (m, 3 H). <sup>13</sup>C

NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 215.7$ , 140.1, 129.2, 128.4, 126.1, 53.7, 43.3, 38.0, 30.4, 29.4, 28.7, 24.3. Anal. Calcd (%) for C<sub>14</sub>H<sub>18</sub>O: C, 83.12; H, 8.97. Found: C, 82.87; H, 8.88. Enantiomeric ratio (er) was determined by HPLC with a Chiracel AD-H column (conditions, Hexane : EtOH = 100 : 1, flow rate = 1 mL / min, 25 °C ), t<sub>r</sub> (minor) = 8.9 min., t<sub>r</sub> (major) = 9.7 min.  $[\alpha]_D^{26.7} = +54.1$ , (*c* 1.0, CHCl<sub>3</sub>) The absolute configuration was established by comparison of the optical rotation to the literature value for (*R*)-(+)-2-benzylcyclohexanone:  $[\alpha]_D = +41.4$  (*c* = 5, MeOH, 88% ee)<sup>17,18</sup>.



The product ((+)-isopropylcycloheptanone) was obtained as a pale yellow oil and 95: 5 er, and <sup>1</sup>H and <sup>13</sup>C NMR was in agreement with the literature<sup>4</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.49$  (td, J = 3.2, 12.8 Hz, 1 H), 2.40-2.30 (m, 1 H), 2.19-2.10 (m, 1 H), 2.00-1.80 (m, 5 H),1.60-1.42 (m, 1 H), 1.42-1.12 (m, 3 H), 0.87 (dd, J = 6.4, 12.8 Hz, 6 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 217.2$ , 59.8, 42.9, 30.6, 30.1, 28.2, 27.8,25.5, 21.1,

19.6. Anal. Calcd (%) for  $C_{10}H_{18}O$ : C, 77.87; H, 11.76. Found: C, 77.81; H, 11.89. Enantiomeric ratio (er) was determined by GC with a CHIRASIL-DEX CB column (conditions, starting temperature: 60 °C [hold 10 min.], rate of temperature increase: 2 °C/min up to 120 °C) after the product was reduced to the corresponding alcohol (major isomer).  $t_r$  (major) = 48.2 min.,  $t_r$  (minor) = 48.4 min.  $[\alpha]_D^{30.0}$  = + 95.5, (*c* 1.0, CHCl<sub>3</sub>).



The product ((*R*)-(+)-2-cyclohexylcycloheptanone) was obtained as a colorless oil and 94: 6 er, and <sup>13</sup>C NMR was in agreement with the literature<sup>12</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.52$  (dt, J = 3.2, 12.4 Hz, 1 H), 2.35-2.28 (m, 1 H), 2.21-2.14 (m, 1 H), 1.98-1.79 (m, 4 H), 1.75-1.40 (m, 7 H), 1.40-1.07 (m, 6 H), 1.17-0.84 (m, 2 H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 217.4$ , 59.3, 42.8, 40.7, 31.4, 30.2, 30.0, 27.9,

27.8, 26.47, 26.44, 25.8. Anal. Calcd (%) for  $C_{13}H_{22}O$ : C, 79.94; H, 11.18. Found: C, 79.98; H, 11.25. Enantiomeric ratio (er) was determined by GC with a CHIRASIL-DEX CB column (conditions, starting temperature: 60 °C [hold 1 min.], rate of temperature increase: 1 °C/min up to 160 °C).  $t_r$  (major) = 85.3 min.,  $t_r$  (minor) = 85.6 min.  $[\alpha]_D^{25.8} = +$  78.3, (*c* 1.0, CHCl<sub>3</sub>) The absolute configuration was established by comparison of the optical rotation to the literature value for (*R*)-(+)-2-cyclohexylcycloheptanone:  $[\alpha]_D^{20} = +$  87.3 (*c* 0.284, CH<sub>2</sub>Cl<sub>2</sub>)<sup>11</sup>.





The product ((*R*)-(+)-2-(4-methylpentyl)cycloheptanone) was obtained as a colorless oil and 92: 8 er, and <sup>1</sup>H NMR was in agreement with the literature<sup>12</sup>. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 2.40-2.32$  (m, 1 H), 2.32-2.18 (m, 2H), 2.13-2.04 (m, 1 H), 2.04-1.92 (m, 1 H), 1.88-1.79 (m, 1 H), 1.79-1.57 (m, 3 H), 1.51 (sep, J = 6.4, Hz, 1 H) 1.42-1.31 (m, 1 H), 1.31-1.08 (m, 5 H), 0.84 (d, J = 6.9 Hz, 6 H). <sup>13</sup>C

NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 213.8, 50.9, 42.1, 39.1, 33.9, 29.7, 28.1, 27.9, 25.0, 24.9, 22.71, 22.66. Anal. Calcd (%) for C<sub>12</sub>H<sub>22</sub>O: C, 79.06; H, 12.16. Found: C, 79.03; H, 12.12. Enantiomeric ratio (er) was determined by GC with a CHIRASIL-DEX CB column (conditions, starting temperature: 60 °C [hold 10 min.], rate of temperature increase: 2 °C/min up to 120 °C). t<sub>r</sub>(major) = 51.1 min., t<sub>r</sub>(minor) = 52.0 min.  $[\alpha]_D^{24.4} = +20.1$ , (*c* 1.0, CHCl<sub>3</sub>). The absolute configuration was established by comparison of the optical rotation to the literature value for (*R*)-(+)-2-(4-Methylpentyl)cycloheptanone:  $[\alpha]_D^{20} = +18.4$  (*c* 3.75, Et<sub>2</sub>O)<sup>12</sup>.



#### 10. Reactivity of enolesters bearing chloroacetyl group and PNP amino acid esters

The pKa values of products (alcohols and acids) are described below (Fig. S1). The pKa value of cyclohexenol is 4.55 point larger than *p*-nitrophenol, although chloroacetic acid is more acidic than *N*-benzoylglycine by 0.75 point. Therefore, enolesters bearing chloroacethyl group seem to be less reactive compared to PNP esters derived from *N*-benzoyl aminoacids.



Fig. S1 Comparison of pKa values between product acids and alcohols (in water)

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S45





S47





S49





















## Expanded figure 1 of HMQC (compound 9)



## Expanded figure 2 of HMQC (compound 9)



## Expanded figure 3 of HMQC (compound 9)



#### **Overall picture of HMBC (compound 9)**



### Expanded figure 1 of HMBC (compound 9)



### Expanded figure 2 of HMBC (compound 9)







#### Expanded figure 4 of HMBC (compound 9)



0.02

0.05

0.04

0.02

0.09

0.07

Y : parts per Million : 13C

0'01

#### Expanded figure 5 of HMBC (compound 9)

5,20

0

0

0

0'I

apangence

1[ms] 0.18[7/m]

Grad 1

abundance

0.1

1.3 1.2 0

7

2.2 2.1 2.0 1.9 1.8 1.7 1.6 1.5

2.5 2.4 2.3

3.0 2.9 2.8 2.7 2.6

( 3.5 3.4 3.3 3.2 3.1 X : parts per Million : 1H

3.6

0'08



#### Expanded figure 6 of HMBC (compound 9)



